Chemoenzymatic Synthesis of an Unnatural Deazaflavin Cofactor That Can Fuel $F_{420}$-Dependent Enzymes

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ABSTRACT: $F_{420}$-dependent enzymes are found in many microorganisms and can catalyze a wide range of redox reactions, including those with some substrates that are otherwise recalcitrant to enzyme-mediated reductions. Unfortunately, the scarceness of the cofactor prevents application of these enzymes in biocatalysis. The best $F_{420}$-producing organism, *Mycobacterium smegmatis*, only produces 1.4 $\mu$mol per liter of culture. Therefore, we synthesized the unnatural cofactor FO-$S^\prime$-phosphate, coined FOP. The FO core-structure was chemically synthesized, and an engineered riboflavin kinase from *Corynebacterium ammoniagenes* (CaRfK) was then used to phosphorylate the $S^\prime$-hydroxyl group. The triple F21H/F85H/A66I CaRfK mutant reached 80% of FO conversion in 12 h. The same enzyme could produce 1 mg (2.5 $\mu$mol) of FOP in 50 mL of reaction volume, which translates to a production of 50 $\mu$mol/L. The activity toward FOP was tested for an enzyme of each of the three main structural classes of $F_{420}$-dependent oxidoreductases. The sugar-6-phosphate dehydrogenase from *Cryptosporangium arvum* (FSD-Cryar), the $F_{420}$-NADPH oxidoreductase from *Thermobifida fusca* (TfuFNO), and the $F_{420}$-dependent reductases from *Mycobacterium haisiacum* (FDR-Mha) all showed activity for FOP. Although the activity for FOP was lower than that for $F_{420}$ with slightly lower $k_{cat}$ and higher $K_m$ values, the catalytic efficiencies were only 2.0, 12.6, and 22.4 times lower for TfuFNO, FSD-Cryar, and FDR-Mha, respectively. Thus, FOP could be a serious alternative for replacing $F_{420}$ and might boost the application of $F_{420}$-dependent enzymes in biocatalysis.

KEYWORDS: deazaflavin, riboflavin kinase, $F_{420}$, biocatalysis, reductase

INTRODUCTION

The naturally occurring cofactor $F_{420}$ was discovered in 1972 in methanogenic archaea where it plays a crucial role in one-carbon catabolism.1 Nowadays, $F_{420}$ is known to be present in a wide range of archaea and bacteria in which it plays an important role in many processes as a redox cofactor.2−5 In several Actinobacteria, for instance, it plays a crucial role in antibiotic synthesis,6−8 as well as aflatoxin degradation,5,9,10 and degradation of other aromatic compounds.11−13 Also, the notorious pathogen *Mycobacterium tuberculosis* has a high abundance of $F_{420}$-dependent proteins.14 This organism, $F_{420}$ is crucial in the regulation of oxidative and nitrosative stress.14−16 Ironically, a series of antitubercular nitroimidazole prodrugs, like pretomanid (PA-824) and delamanid (OPC-67683), are specifically activated by a $F_{420}$-dependent reductase in vivo, releasing toxic NO.17−19

Structurally, the 7,8-dimethyl-8-hydroxy-5-deazariboflavin catalytic core of $F_{420}$, called FO, is analogous to that of riboflavin20 (see Figure 1). Its chemistry, however, resembles more that of nicotinamide dinucleotide (NAD($P^+$)), as $F_{420}$ can only perform two-electron hydride transfers and is hardly reactive toward molecular oxygen. Its reduction potential of $-360$ mV is lower than that of NAD($P^+$) and the riboflavin derived cofactors flavin mononucleotide (FMN) and flavin adenine dinucleotide (FAD), which are $-320$ and $-220$ mV, respectively.21 Apart from the FO core, the rest of the structure is very different than that of the nicotinamide and flavin cofactors. The ribityl tail of FO is extended with a $S^\prime$-hydroxyl group, forming $F_{420}$-0, and this is in turn elongated by...
a poly-γ-glutamyl tail.\textsuperscript{20} The length of the poly-γ-glutamyl tail depends on the organism and varies from 2 to 9 monomers.\textsuperscript{3,22} Interestingly, the larger part of the poly-γ-glutamyl tail is not bound to the enzyme, as was seen in several crystal structures and modeled protein structures.\textsuperscript{5,23–28} Ney et al.,\textsuperscript{3} however, showed that electrostatic interactions of the poly-γ-glutamyl tail with enzymes of the split β-barrel like fold flavin/deazaflavin oxidoreductase (FDOR) and TIM barrel fold luciferase-like hydride transferase (LLHT) families influence the binding affinity. It seems that a longer polyglutamyl tail results in a higher binding affinity (lower $K_d$ and $K_{cat}$), which lowers the catalytic turnover ($k_{cat}$), probably because of slower cofactor exchange rates.

The uniquely low redox potential of the cofactor makes it an interesting candidate for the use in biocatalysis.\textsuperscript{29,30} F420-dependent enzymes can reduce various physiologically important heterocyclic enones, unsaturated esters, and imines, which are inert to other enzymes.\textsuperscript{5,10,31,32} Recently, it was shown that F430-dependent reductases (FDR), part of the split β-barrel fold FDORs, can reduce αβ-unsaturated ketones and aldehydes in an enantio- and regioselective fashion and, interestingly, yielding the opposite enantiomer as would be formed by the well-studied FMN-dependent old yellow enzyme-type reductases.\textsuperscript{33} Also F420-dependent enantioselective secondary alcohol dehydrogenases were characterized.\textsuperscript{23,34} These enzymes could also be used as cofactor recycling systems to supply reductases with reduced F420. The well-studied F420/NADPH oxidoreductases\textsuperscript{26,35–39} and sugar-6-phosphate dehydrogenases\textsuperscript{27,40–44} could also be used as recycling systems. These enzymes form a biocatalytic toolbox which is anticipated to expand, as many genomes are predicted to accommodate genes for F420-dependent enzymes that have yet to be characterized.\textsuperscript{34}

The main bottleneck in the application of F420-dependent enzymes thus far is the limited availability of the cofactor. Many of the organisms that produce F420 are hard to culture or yet to be characterized.\textsuperscript{4,29} Many of these organisms was, until recently, impossible because of some missing links in the F420-biosynthesis pathway. The recent elucidation of the QIAquick PCR Purification Kit (Qiagen, Valencia, CA, U.S.A.).

**Materials and Methods**

Reagents and chemicals were purchased from Sigma-Aldrich (St. Louis, MO, U.S.A.) unless indicated otherwise. Mutagenic primers were also ordered at Sigma-Aldrich. Ligase, and restriction endonucleases were obtained from New England Biolabs (NEB, Ipswich, MA, U.S.A.). PfuUltra Hotstart PCR Mastermix (Agilent Technologies) was used for mutagenic PCR (QuickChange). Plasmid DNA was isolated using the QIAprep Miniprep Kit, and PCR products were purified with the QIAquick PCR Purification Kit (Qiagen, Valencia, CA, U.S.A.).

**F420 Production.** F420 was isolated from *Mycobacterium smegmatis* as described by Bashiri et al. and Isabelle et al.\textsuperscript{45,46} The production strain *M. smegmatis* mc² 4517 was a kind gift from Dr. G. Bashiri from the University of Auckland, New Zealand.

**FO and FO-7-Methyl Synthesis.** FO was synthesized using the approach described by Hossain et al.\textsuperscript{48} with a modification for the reductive amination procedure. These two steps are done in one step via reductive amination with sodium cyanoborohydride, instead of synthesis of the mixture of anomers and then reduction of them. FO-7-Me was synthesized using the same modified procedure as for synthesis of FO. Detailed procedures and physical data can be found in the Supporting Information.

**Gene Cloning and Mutagenesis of *C. ammoniagenes* Riboflavin Kinase.** The *C. ammoniagenes* riboflavin kinase gene (CaRFK) was ordered at GenScript (Piscataway, NJ, U.S.A.), codon optimized for *E. coli*. It was composed of the C-terminal kinase domain of the FAD synthetase gene, *ribF* (NCBI#: D37967.1), previously described by Iamurri et al.\textsuperscript{39} The nucleotide sequence and protein sequence are shown in Figures S1 and S2, respectively. The gene was cloned into a pBAD/Myc-His vector (Invitrogen, Thermo-Fisher) using restriction sites NdeI and HindIII, following standard cloning procedures.\textsuperscript{50} Site-directed mutagenesis was performed on the riboflavin kinase gene with the use of mutagenic primers, degenerate at a chosen codon, using the QuickChange mutagenesis kit (Stratagene), following the procedure of the manufacturer. Primers were designed with the Agilent QuikChange primer design tool (http://www.genomics.agilent.com/primerDesignProgram.jsp). The used primers are listed in Table S1. Sequencing was performed at GATC (Constance, Germany). The plasmids were transformed into calcium chloride chemically competent *E. coli* NEB 10-beta (New England Biolabs Ipswich, MA, U.S.A.) for amplification and protein expression, using standard protocols.\textsuperscript{50}

**Expression and Purification of *C. ammoniagenes* Riboflavin Kinase.** An *E. coli* NEB 10-beta overnight culture in Terrific broth (TB), supplemented with 50 μg mL⁻¹ ampicillin, was grown at 37 °C, 135 rpm. The overnight culture was diluted a hundred times in 200 mL of fresh TB with 50 μg mL⁻¹ ampicillin in a 500 mL Erlenmeyer flask. This was grown at 37 °C, 135 rpm until the OD₆₀₀ reached ~0.5, at which the culture was induced with 0.2% L-arabinose and further grown at 17 °C for 36 h. Cells were harvested by centrifugation at 4000g for 20 min at 4 °C. The cell pellets were stored at −20 °C until purification. Cell pellets were resuspended in about 10 mL of 50 mM Tris/HCl pH 8.0, 200 mM NaCl, 10 mM imidazole, 1 mM β-mercaptoethanol, and
cComplete mini EDTA-free Protease Inhibitor Cocktail. The cells were lysed by sonication, using a Sonics Vibra-Cell VCX 130 sonicator with a 3 mm stepped microtip (5 s on, 10 s off, 70% amplitude, 7.5 min). Cell debris were pelleted by centrifugation at 8000g for 40 min at 4 °C. The clear supernatant was incubated on 2 mL of Ni-Sepharose High Performance (GE Healthcare, Eindhoven, The Netherlands) for 12 h at 4 °C, with gentle shaking. The column was washed extensively with 3 column volumes of 50 mM Tris/HCl pH 8.0, 200 mM NaCl, 20 mM imidazole, and the protein was eluted with 50 mM Tris/HCl pH 8.0, 200 mM NaCl, 300 mM imidazole. The eluted protein was then desalted and concentrated using Amicon Ultra centrifugal filter units with a 3 kDa molecular weight cut-off, exchanging the buffer with 50 mM Tris/HCl pH 8.0, 200 mM NaCl. The protein was flash-frozen in liquid nitrogen and stored at −80 °C until further use. Purity was checked with SDS-PAGE analysis, and protein concentrations were measured by Bradford analysis, using the standard protocols.

Expression and Purification of F420-Dependent Enzymes. Genes, plasmids, and host strains were already in the collection of this lab from earlier studies. T. fusca F420:NADPH oxidoreductase (TfuFNO) was expressed and purified as described by Kumar and Nguyen et al.26 R. jostii RHA1 F420-dependence glucose-6-phosphate dehydrogenase (RHA1-FGD) was expressed and purified as described by Nguyen et al.27 The sugar-6-phosphate dehydrogenase from C. arvum (FSD-Cyr) was expressed and purified as described by Mascetti and Kumar et al.41 The F420-dependent reductases from M. hassiacum (FDR-Mha) and R. jostii RHA1 (FDR-RHA1) were expressed and purified as described by Mathew and Trajkovic et al.33

Riboflavin Kinase Activity Assay and HPLC Analysis. The activity of wild-type CaRFK and mutant enzymes toward riboflavin and FO was measured in conversion experiments. Conversion mixtures contained 1 μM enzyme and 50 μM riboflavin, FO-7-methyl (FO-7-Me) or FO in 50 mM Tris/HCl pH 8, 100 mM MgCl₂, 10 mM ATP. The reaction mixtures contained a total volume of 0.5 mL were incubated at room temperature or 37 °C for 12 h. Samples were taken either at certain intervals (0, 5, 10, 15, 30 min) or after 12 h. Conversions were measured by either high-performance liquid chromatography (HPLC) or thin layer chromatography (TLC). The TLC method was previously described by Iamurri et al.49 The HPLC method was a modified version of that of Iamurri et al. The reactions were quenched with 100% formic acid (FA), 1:5 FA:sample, incubating on ice for 5 min. Then, the samples were spun down at 8000g in a table top centrifuge and applied to a Revellers C18-WP Flash cartridge column. FOP was eluted with deionized water, and purity was verified by HPLC, as described above. The product was concentrated by water evaporation under reduced pressure with a rotary evaporator. FOP was either kept at −20 °C for long-term storage or at 4 °C for short-term storage. Obtained FOP was confirmed by HRMS (result is shown in Supporting Information).

Steady-State Activity Assays for Selected F420-Dependent Enzymes with FO, FOP, and F420. The Michaelis–Menten kinetic parameters for TfuFNO with FO and FOP were obtained by the spectrophotometric assay as described by Kumar and Nguyen et al.26 In short, the measurements were performed at 25 °C by adding 25–50 μM enzyme to 50 mM KPi, pH 6.0, with a constant NADPH concentration of 250 μM and varying concentrations of FO and FOP between 0.625 and 50 μM. The activity of RHA1-FGD and FSD-Cyr with FO, F420 and FOP was obtained by the spectrophotometric assay as described by Nguyen et al.27 and Mascetti and Kumar et al.,41 respectively. Glucose-6-phosphate was used as substrate at a constant concentration of 20 mM. The concentration of FO, FOP, and F420 was varied between 1.25 and 50 μM in the appropriate Tris/HCl-based buffers for each enzyme at 25 °C.27,41

The absorbance at 400 nm was followed in time for all the experiments, and observed slopes (kobs) were calculated with ε400 (F420) = 25.7 mM−1 cm−1. All experiments were performed in triplicates. The kobs values were plotted against de FO/FOP/F420 concentration, and the data was fitted to the Michaelis–Menten (eq 1) or Hill equation (eq 2) by nonlinear regression, using GraphPad Prism v. 6.0 (GraphPad Software Inc., La Jolla, CA, U.S.A.).

\[
k_{\text{obs}} = \frac{k_{\text{cat}}[S]}{k_{m} + [S]} \quad (1)
\]

\[
k_{\text{obs}} = \frac{k_{\text{cat}}^h[S]^h}{k_{\text{h}}[S]^{h}} \quad (2)
\]

The kinetic parameters for the F420-dependent reductase FDR-Mha were obtained as follows: FOPH₂ and F420H₂ were prepared by incubating 500 μM FOP or F420 with 10 μM FSD-Cyr and 20 mM glucose-6-phosphate in 50 mM Tris/HCl, pH 8.0, until the yellow color disappeared. Then, the mixture was passed through an Amicon Ultra 0.5 mL centrifugal filter, 10 kDa molecular weight cutoff. The filtrate, containing 500 μM FOPH₂ or F420H₂, was then immediately used for a spectrophotometric assay. The assay mixture contained 0.1–1 μM FDR-Mha, 1.0 mM 2,6,6-trimethyl-2-cyclohexene-1,4-dione, and various concentrations of FOPH₂ and F420H₂ between 1.25 and 50 μM in 50 mM Tris/HCl, pH 8.0. The increase in absorbance at 400 nm was measured in time over several minutes. The kobs values were calculated using ε400 (F420) = 25.7 mM−1 cm−1. All experiments were performed in triplicates at 25 °C. Kinetic data was analyzed by nonlinear regression and fitted to the Hill equation for cooperative binding (eq 2), using GraphPad Prism v. 6.0 (GraphPad Software Inc., La Jolla, CA, U.S.A.).

Conversion Experiments with F420-Dependent Reductases from M. hassiacum and R. jostii RHA1 with FO, FOP, and F420. The reaction mixture contained 400 μL of 50 mM Tris/HCl pH 8.0 supplemented with 1.0 mM cinnamaldehyde, 20 μM FO/FOP/F420, 1.0 μM TfuFNO, 10

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RESULTS

Engineering of C. ammoniagenes Riboflavin Kinase toward Activity on FO. The C-terminal riboflavin kinase domain (CaRFK) of the FAD synthetase (RibF) from C. ammoniagenes, previously described by Iamurri et al. and Herguedas et al., was chosen for the biocatalytic phosphorylation of FO. The truncated ribF gene (NCBI#: D37967.1), CaRFK, was ordered, codon optimized for E. coli and was transformed into E. coli NEB 10-beta as a pBAD-CuRFK construct, equipped with a C-terminal 6x histidine tag. Purification of the respective protein, CaRFK, was achieved by table-top nickel affinity chromatography, yielding about 20 mg L\(^{-1}\) pure CaRFK.

Conversion experiments with 50 \(\mu\)L of FO and 1–50 \(\mu\)M CaRFK, analyzed by reverse-phase HPLC after 12 and 48 h at 20 or 37 \(^\circ\)C, showed that the wild-type enzyme had no detectable activity toward FO. Therefore, structure-guided site-directed mutagenesis was performed. The crystal structure of the C-terminal riboflavin kinase domain of RibF, PDB ID 5A89, was used to identify suitable sites for mutagenesis. Polar amino acids were introduced at positions 21, 85, and 122 and the mutation T23S, both in the same binding pocket facing the 8-hydroxyl group of FO, did not increase the activity any further.

Chemoenzymatic Synthesis of FO-7-Me-P and FOP. Incubating 50 \(\mu\)M FO-7-Me with 10 \(\mu\)M F21H/F85H CaRFK mutant or FO with 10 \(\mu\)M F21H/F85H/A66I CaRFK mutant in a volume of 50 mL resulted in full conversion of FO-7-Me to FO-7-Me-P and FO to FOP within 24 h at 37 \(^\circ\)C. The product could be purified from the other reaction components with the aid of a preparative reverse-phase liquid chromatography and was concentrated by rotary evaporation under reduced pressure. The yield of 50 mL of reaction volume was about 1 mg (2.5 \(\mu\)mol) of FOP and FO-7-Me-P, in a final concentration of about 400 \(\mu\)M (measured by absorbance at 400 nm, \(\varepsilon_{400} = 25.7\) mM\(^{-1}\) cm\(^{-1}\)). Obtained FO-7-Me-P and FOP were confirmed by HRMS (see Supporting Information).

Purification of F420-Dependent Oxidoreductases. The F420-NADPH oxidoreductase from T. fusca (TfuFNO) was expressed and purified as described by Kumar and Nguyen et al. with a yield of approximately 140 mg L\(^{-1}\) pure protein. The glucose-6-phosphate dehydrogenase from R. jostii RHA1 (RHA1-FGD) and the sugar-6-phosphate dehydrogenase from C. arvum (FSD-Cryar) were expressed and purified as described by Nguyen et al. and Mascotti and Kumar et al., respectively. Yields of purified enzyme were similar as described by the papers mentioned above. Similar results were also obtained for the expression and purification of the F420-dependent reductases from M. hassiacum (FDR-Mha) and R. jostii RHA1 (FDR-RHA1), as described by Mathew and Trajkovic et al.

Steady-State Kinetics Using FOP as Alternative Cofactor. The activity of the F420-dependent enzymes TfuFNO, RHA1-FGD, FSD-Cryar, and FDR-Mha toward FO, FOP, and F420 was measured spectrophotometrically. Steady-state kinetic parameters were measured by varying the concentration of the coenzymes FO, FOP, and F420 while...
keeping the other substrates at constant, saturating concentrations. The slopes of absorbance decrease—or increase in the case of FDR-Mha—at 400 nm were measured, and the observed rates \((k_{\text{cat}})\) were calculated using \(A_{400} = 25.7 \text{ mM}^{-1} \text{ cm}^{-1}\). The observed rates were plotted against the cofactor concentration and fitted to the Michaelis–Menten or the Hill model for positive cooperativity (eqs 1 and 2, respectively).

TfuFNO showed activity with all three tested deazaflavins (FO, FOP, and F420). The kinetic data with FO and FOP did not fit well to the Michaelis–Menten kinetic model but could be fitted to a Hill plot for cooperative substrate binding kinetics (Figures S4 and S5). This positive cooperativity can be explained by the dimeric structure of the enzyme. Cooperativity was also observed for the FNO from *Archeoglobus fulgidus*, using NADPH and FO, as reported by Le et al.\(^{25}\) Strangely, positive cooperativity was not seen with F420 as a coenzyme, as discussed by Kumar and Nguyen et al.\(^{26}\) The kinetic parameters for FO and FOP, however, are similar to that of F420 (see Table 1). Thus, FO, FOP and F420 are all equally well used as coenzymes in NADPH oxidation by TfuFNO. The crystal structure of TfuFNO (PDB ID 5N2I),\(^{26}\) superimposed with the crystal structure of the F420-bound FNO homologue from *A. fulgidus* (1JAY),\(^{15,39}\) shows that only the FO part of F420 is bound and that the rest of the molecule is actually located outside of the enzyme. Therefore, the phosphate of FOP or phospho-L-lactyl and poly-γ-glutamyl tail of F420 have only minor contributions to the binding affinity.

It was found that RHA1-FGD shows no activity with FO or FOP, while it has a \(k_{\text{cat}}\) of 17 s\(^{-1}\) and \(K_m\) of 3.8 μM for F420 \(^{27}\). The crystal structure of RHA1-FGD (PDB ID SLXE),\(^{27}\) superimposed with the F420-bound crystal structure of the *M. tuberculosis* homologue (PDB ID 3Y4B),\(^{23}\) shows that the first glutamate moiety of the F420 poly-γ-glutamyl tail forms hydrogen bonds with the protein backbone. This could be the reason for being unable to use FO or FOP as coenzyme, which are lacking the glutamyl tail. FSD-Cryar, however, with \(S7\)% sequence identity to RHA1-FGD, did show detectable activity toward FOP. The \(k_{\text{obs}}\) data fitted well to a Michaelis–Menten curve and gave a \(k_{\text{cat}}\) of 1.3 s\(^{-1}\) and \(K_m\) of 7.0 μM for FOP (Figures S6 and S7, Table 1). Both values are lower than the kinetic parameters measured for the native cofactor F420 being 33 s\(^{-1}\) and 13.9 μM for the \(k_{\text{cat}}\) and the \(K_m\) respectively (Table 1). Although the \(K_m\) values are in the same range (\(K_m\_\text{FO} \approx 2\) times lower than \(K_m\_\text{F420}\)), the \(k_{\text{cat}}\) value for F420 is 25 times higher than that for FOP. The catalytic efficiency for FOP, however, is only 1 order of magnitude lower than that when using F420. The measured \(k_{\text{obs}}\) values for different concentrations of FO could not be fitted with the formula for the Michaelis–Menten model. A good fit could be obtained with a sigmoidal function, but the plateau was never reached with FO concentrations up to 80 μM. Higher concentrations could not be tested because of the solubility of FO in buffer. Although the data could not be used to calculate the kinetic parameters of FSD-Cryar with FO, it is safe to say that the \(K_m\) must be significantly higher than that of FOP and F420. The \(k_{\text{cat}}\) at a concentration of 80 μM FO, the highest concentration tested, is about 20% of the \(k_{\text{cat}}\) with FOP (see Figure S8).

The kinetic parameters could also be established for FDR-Mha with both FOP and F420 as coenzymes. But the kinetic parameters for FO, however, could not be measured. The reduced form of this deazaflavin could not be obtained as it could not be reduced by FSD-Cryar or RHA1-FGD. TfuFNO, which can reduce FO, could not be used as the necessary excess of NADPH would interfere with the FO absorbance at 400 nm. The \(k_{\text{obs}}\) data for both F420 and FOP fitted well to a sigmoidal Hill-curve (eq 2; Figures S9 and S10). This could mean positive cooperativity or the fact that the lower concentrations of the coenzymes were in the same range as the enzyme concentration, so not being present in saturating conditions and, thus, not fitting to the Michaelis–Menten model. The same sigmoidal behavior was also seen for the various concentrations of F420.\(^{5}\) The asymptotes to the plateau values of the curves still give a good description of the \(k_{\text{cat}}\) and the \(K_m\) still gives an accurate estimation for coenzyme specificity. From Table 1, it can be seen that FOP has a somewhat lower \(k_{\text{cat}}\) and higher \(K_m\) value than F420, but they are still in the same range.

### Conversion Experiments with FDR-Mha and FDR-RHA1

Since the catalytic parameters could not be determined spectrophotometrically for FDR-Mha and FO, conversion experiments were used to compare the activity of FDR-Mha and FDR-RHA1 with FO, FOP, and F420. The conversion of cinnamaldehyde was measured by reverse-phase HPLC after 3 h of incubation for all three coenzymes at concentration of 20 μM. The results show that the conversion of cinnamaldehyde is around 70% when using F420 for both FDR-Mha and FDR-RHA1. The conversion when using FOP is close to 40%, which is in line with the spectrophotometric experiments. The conversion of cinnamaldehyde with FO as a coenzyme is lower than that of FOP by about 10 to 18 percentage points. These results show that the apparent activity follows the trend F420 > FOP > FO (see Figure 4).

### Table 1. Steady-State Kinetic Parameters with Three Different Deazaflavin Cofactors

<table>
<thead>
<tr>
<th>enzyme</th>
<th>cofactor</th>
<th>(k_{\text{cat}}) (s(^{-1}))</th>
<th>(K_m) or (K_m) (μM)</th>
<th>(k_{\text{cat}}/K_m) (M(^{-1}) s(^{-1}))</th>
<th>(k_{\text{cat}}/K_m)</th>
<th>(F_420/F_{\text{FOP}})</th>
</tr>
</thead>
<tbody>
<tr>
<td>TfuFNO</td>
<td>F420</td>
<td>3.3(^{1,26})</td>
<td>2.0(^{1})</td>
<td>1.7 × 10(^6)</td>
<td>2.0</td>
<td></td>
</tr>
<tr>
<td></td>
<td>FOP</td>
<td>3.3</td>
<td>4.0(^{1,3}) (1.8)</td>
<td>8.3 × 10(^5)</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>FO</td>
<td>2.2</td>
<td>4.8(^{1,3}) (2.9)</td>
<td>4.6 × 10(^5)</td>
<td></td>
<td></td>
</tr>
<tr>
<td>FSD-Cryar</td>
<td>F420</td>
<td>33.0</td>
<td>13.6</td>
<td>2.4 × 10(^6)</td>
<td>12.6</td>
<td></td>
</tr>
<tr>
<td></td>
<td>FOP</td>
<td>1.3</td>
<td>7.0</td>
<td>1.9 × 10(^5)</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>FO</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td></td>
<td></td>
</tr>
<tr>
<td>FDR-Mha</td>
<td>F420</td>
<td>1.8 × 10(^{-2})</td>
<td>12.1(^{1,3}) (4.4)</td>
<td>1.5 × 10(^4)</td>
<td>22.4</td>
<td></td>
</tr>
<tr>
<td></td>
<td>FOP</td>
<td>1.3 × 10(^{-2})</td>
<td>19.3(^{1,3}) (2.9)</td>
<td>6.7 × 10(^3)</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>FO</td>
<td>-</td>
<td>-</td>
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</tbody>
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\(^{1}\)Values described by Kumar et al.\(^{26}\) \(^{2}\)\(K_{\text{cat}}\) values, according to eq 2, Hill coefficients (h) are given in between brackets. Errors are within a 10% range.
Enzymes that utilize F420 could become a biotechnological tool of importance in the near future. The very low redox potential of the cofactor can be utilized for the bioconversion of compounds that would otherwise be recalcitrant to enzymes. The now known F420-dependent reductases and cofactor recycling systems harbor interesting activities for biocatalysis. Genome studies revealed that many bacteria and archaea contain putative F420-dependent enzymes that yet have to be characterized and could potentially yield new biocatalysts with novel interesting properties.

A bottleneck for application of F420-dependent enzymes is the low availability of the cofactor, as many F420-producing organisms are slow-growing and hard to culture, resulting in low yields. Furthermore, the complex structure of the cofactor prevents straightforward chemical synthesis methods. The recent and remarkable efforts by Bashiri et al. to heterologously produce F420 in E. coli unfortunately still resulted in similar low yields as with M. smegmatis. This inspired us to synthesize a F420 analogue that could replace F420.

The drawback of using riboflavin kinase is its low stability, which prevents the use of cosolvents. Since the solubility of FO and other riboflavin analogues in water is quite low, relatively large reaction volumes have to be used for the production of large amounts of phosphorylated compounds. A more stable RKF could be a solution to this problem, if one would like to produce FOP or other cofactors on plant scale. Still, a 50 mL reaction resulted in the production of about 1 mg (2.5 μmol) FOP within 24 h, which translates to a production of 50 μmol/L. Thus, far more than the F420 yield from M. smegmatis, being 1.4 μmol/L, is obtained during several days.

The activity of an F420-dependent oxidoreductase from each structural class with FO, FOP, and F420 was measured. Steady-state parameters could be determined with FOP and F420 for a member of each class, spectrophotometrically. Experiments with the F420-NADPH oxidoreductase from T. fusca, TfuFNO, a member of the Rossmann fold structural class, showed that the kinetic parameters for FO, FOP, and F420 are in the same range. The K_m values of FO and FOP are slightly higher, resulting in a slightly lower catalytic efficiency that is 3.7 (FO) or 2 (FOP) times lower than that for F420. The similarity in kinetic parameters can be explained by the fact that only the FO core is bound to the enzyme, as can be seen in crystal structures.

Members from the TIM barrel fold class showed a significant decrease in activity when FO and FOP were used as coenzymes. The F420-dependent glucose-6-phosphate dehydrogenase from R. jostii RHA1 shows no detectable activity at all for the alternative deazaflavin cofactors. The sugar-6-phosphate dehydrogenase from C. arvn has activity toward FOP. Although the K_m values for both FOP and F420 were lower than the kinetic parameters for F420, the enzyme is still specific toward FOP.
are similar, the $k_{\text{cat}}$ for F$_{420}$ is significantly higher than that of FOP. Still, the catalytic efficiency with FOP is only 1 order of magnitude lower, meaning that the activity with FOP is significant and that FOP could be used as an alternative non-natural cofactor for this enzyme. An even lower activity was found for FO, indicating that the phosphate moiety of the F$_{420}$ cofactor is important for recognition and/or productive coenzyme binding.

The kinetic parameters for both FDR-MH1 and FDR-Mha, both members of the split $\beta$-barrel fold for F$_{420}$ and FOP revealed that the $K_m$ and $k_{\text{cat}}$ values are in the same order of magnitude, although the catalytic efficiency is 22 times higher for F$_{420}$ than for FOP. It shows that members of this group of F$_{420}$-dependent enzymes, dedicated to perform reductions, also can utilize the non-deazaflavin cofactor that lacks the lactyl-poly glutamyl moiety. No steady-state kinetic parameters could be obtained with the reduced form of FO because the sugar-6-phosphate dehydrogenases could not be used to prepare reduced FO. We decided to perform comparative conversion experiments with cinnamaldehyde as substrate to probe the efficiency of all three deazaflavin cofactors. Here we could see that cinnamaldehyde conversion with FOP is somewhat less effective than with F$_{420}$, which is in line with the results from the spectrophotometric assay. The conversions with FO are again lower than that with FOP but in the same range. It shows that all three deazaflavins are accepted by the tested reductases, suggesting that for catalysis only the FO part of the F$_{420}$ cofactor is essential.

The experiments have shown that F$_{420}$-dependent enzymes of all tested structural classes are active on the non-natural, synthetic deazaflavin cofactor FOP and that the activity is typically higher when compared with its precursor, FO. It seems that in all cases, the phosphate group is important for a higher binding affinity, as seen by a lower $K_m$, and for a higher catalytic turnover, as seen by a higher $k_{\text{cat}}$. When comparing F$_{420}$ and FOP activities the presence of a poly-$\gamma$-glutamyl tail seems to be important for higher cofactor affinities and higher catalytic activity. Interestingly, Ney et al. discovered that the presence of a long poly-$\gamma$-glutamyl tail of 5 to 8 glutamate residues indeed lowered the $K_m$ as compared with a shorter tail of two residues, because of favorable electrostatic interactions between tail and enzyme, but also lowered the $k_{\text{cat}}$ and vice versa. The effect of higher $k_{\text{cat}}$ with lower cofactor affinity can be explained by higher cofactor exchange rates of enzymes at saturating conditions. Surprisingly, the same effect on $k_{\text{cat}}$ and $K_m$ is not seen when the poly-$\gamma$-glutamyl tail is completely missing. Perhaps, the properties of the more substituted phosphate diester in F$_{420}$ are more favorable for enzyme binding than that of the phosphate monoester in FOP.

The findings of this research show that FOP could also be used as an alternative deazaflavin cofactor in vivo. With having an engineered FO kinase, the introduction of a biosynthetic route toward FOP is feasible. A great advantage of FOP over FO is that it retains inside a cell because of the charged phosphate group, similar to the conversion of riboflavin into FMN. Also, the solubility is enhanced by the addition of a phosphate group. Although the activity for FOP is lower than that for F$_{420}$ it is still sufficient enough for the use as a cofactor in biocatalysis, as can be seen by the relatively small differences in catalytic efficiency. The ease of production and higher product yields of the cofactor could make up for the decrease in activity when used as coenzyme in F$_{420}$-dependent enzyme reactions. Therefore, FOP could be the key to successful implementation of F$_{420}$-dependent enzymes as biocatalytic tools in industry.

**ASSOCIATED CONTENT**

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acscatal.9b01506.

Protein and nucleotide sequence of cinnamaldehyde CaRFK, as well as the mutantic primers used to create point mutations in the CaRFK gene; synthesis of FO and FOP-7-Me and all the NMR and HRMS spectra for product and intermediate characterization and identification; and all graphs from steady-state kinetic analysis of the F$_{420}$-dependent enzymes with F$_{420}$/FOP/FO (PDF)

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**Notes**

The authors declare no competing financial interest.

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