Catalytic Direct Cross-Coupling of Organolithium Compounds with Aryl Chlorides

Valentín Hornillos, Massimo Giannerini, Carlos Vila, Martín Fañanás-Mastral,* and Ben L. Feringa*

Stratingh Institute for Chemistry, University of Groningen, Nijenborgh 4, 9747 AG Groningen, The Netherlands

b.l.feringa@rug.nl; m.fananas.mastral@rug.nl

Received September 4, 2013

ABSTRACT

Palladium-catalyzed direct cross-coupling of aryl chlorides with a wide range of (hetero)aryl lithium compounds is reported. The use of Pd-PEPPSI-IPent or Pd₂(dba)₃/XPhos as the catalyst allows for the preparation of biaryl and heterobiaryl compounds in high yields under mild conditions (room temperature to 40 °C) with short reaction times.

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The development of new catalytic methodologies for C–C bond formation continues to be a major challenge in organic synthesis.¹ Cross-coupling reactions, in particular Pd-catalyzed processes, are among the most important current methods for C–C bond formation.¹ The well-established Stille (with organotin as the nucleophile),² Suzuki–Miyaura (organoboron),³ Negishi (organozinc),⁴ Hiyama–Denmark (organosilicon),⁵ or Kumada (organomagnesium)⁶ reactions are widely employed for this transformation with numerous applications in disciplines varying from material science to natural products synthesis, asymmetric catalysis, and medicinal chemistry.¹,⁷

Organolithium compounds are among the most versatile and widely used reagents in organic synthesis.⁸ Their use is also well documented for the preparation of softer organometallic nucleophiles including Sn, Si, or B reagents.⁹ However, their direct use in catalytic cross-couplings, which would eliminate this additional transformation, has been largely prohibited due to the high reactivity.

usually accompanied by the lack of selectivity.\textsuperscript{10} Our group has recently described the direct use of organolithium reagents in the Pd-catalyzed cross-coupling of a wide variety of aryl and alkenyl bromides.\textsuperscript{11} By the use of toluene as a solvent and \( \text{P}^3(\text{Bu})_3 \) as ligand and by fine-tuning of the reaction conditions, the high reactivity of the organolithium reagents was controlled, efficient transmetalation was achieved, and high selectivity and good yields were obtained, avoiding the notorious lithium halogen exchange and homocoupling side reactions. Nonetheless, the coupling of organolithium reagents with the corresponding aryl chlorides remains challenging. Aryl chlorides are generally more desirable substrates than their corresponding bromide and iodide counterparts taking advantage of low cost and availability.\textsuperscript{12} However, their low reactivity has traditionally made these substrates inconvenient to use in cross-coupling methodologies.\textsuperscript{13} By the use of a silicon-based transfer agent, see: (c) Smith, A. B., III; Hoye, A. T.; Martinez-Solorio, D.; Kim, W.; Tong, R. J. Am. Chem. Soc. 2012, 134, 1229. (d) Nguyen, M. H.; Smith, A. B. III; Org. Lett. 2013, 15, 4268.

We surmised that the development of new cross-coupling methodology which combines both, cheap and easy accessible organolithium reagents and aryl chlorides is highly desirable. The anticipated process would drastically reduce the amount of byproducts, the light and nontoxic LiCl being the only stoichiometric reaction waste. Herein, we report that the use of the commercially available Pd-PEPPSI-IPent or Pd\(_2\)(dba)\(_3\)/XPhos catalysts allows the selective cross-coupling of (hetero)aryllithium compounds with aryl chlorides in high yields under mild conditions (rt to 40 °C) and short reaction times (40 min to 4 h).

We started this study with the reaction between phenyllithium and 2-chloronaphthalene 1a. Under the optimized conditions for the cross-coupling of organolithium reagents with aryl bromides (Pd\(_2\)(dba)\(_3\)/P\(^3(\text{Bu})_3\)),\textsuperscript{11} the desired product 2a was obtained in the presence of a large amount of homocoupling side product 4 (Table 1, entry 1).

Table 1. Screening of Different Ligands (see also Supporting Information, Table S1)

<table>
<thead>
<tr>
<th>entry</th>
<th>[Pd]</th>
<th>ligand</th>
<th>conv (%)</th>
<th>2a:3:4:5b</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Pd(_2)(dba)(_3)</td>
<td>L1, P(Bu)_3</td>
<td>78</td>
<td>46:3:26:2</td>
</tr>
<tr>
<td>2</td>
<td>Pd(_2)(dba)(_3)</td>
<td>L2, P(Cy)_3</td>
<td>90</td>
<td>49:2:20:19</td>
</tr>
<tr>
<td>3</td>
<td>Pd(_2)(dba)(_3)</td>
<td>L3, XPhos</td>
<td>full</td>
<td>99:1:1:0</td>
</tr>
<tr>
<td>4</td>
<td>Pd-PEPPSI-IPent</td>
<td>full</td>
<td>94:3:2:0</td>
<td></td>
</tr>
<tr>
<td>5\textsuperscript{d}</td>
<td>Pd-PEPPSI-IPent</td>
<td>full</td>
<td>97:3:1:0</td>
<td></td>
</tr>
</tbody>
</table>

\( ^{a} \) Conditions: PhLi (0.45 mmol, 1.8 M solution in dibutyl ether diluted with THF to a final concentration of 0.6 M) was added (1 mL) to a solution of 2-chloronaphthalene (0.3 mmol) in toluene (2 mL) at rt. 

\( ^{b} \) Ratios determined by GC analysis. 

\( ^{d} \) 2a:3:4:5 ratios determined by GC analysis. 

The use of PC\(_3\) was detrimental for the selectivity and 1-phenylnaphthalene 5 was also formed indicating the formation of a benzyne intermediate via 1,2-elimination promoted by the organolithium reagent (entry 2).\textsuperscript{16} To our delight, when XPhos (L3) was used in combination with Pd\(_2\)(dba)\(_3\) (2.5 mol %) the cross-coupled product 2a was obtained with excellent selectivity (> 99%), avoiding dehalogenation (< 1%) and inhibiting the formation of the homocoupling or isomerized side products (< 1%, entry 3). The use of other sterically hindered phosphines resulted in lower selectivity with incomplete conversion (see Supporting Information (SI), Table S1). We also evaluated catalysts based on NHC ligands and observed that the air stable Pd-PEPPSI-IPent, introduced by Organ,\textsuperscript{15c} also displayed high reactivity and selectivity (entry 4). In this...
case, slower addition of PhLi (0.5 mL/h) and a higher concentration were found to be beneficial for an increase of the selectivity (entry 5). It should be noted that, for both catalytic systems, the reaction proceeds at rt and is finished once the addition of the lithium reagent is completed (40 min to 1 h) providing 2a in 88% and 93% yield, respectively.

Having in hand these two optimized catalytic systems (A based on Pd-PEPPSI-IPent and B using Pd2(dba)3/XPhos), we examined the cross-coupling between different organolithium reagents and aryl chlorides (Scheme 1).

Scheme 1. Pd-Catalyzed Cross-Coupling of Aryl Lithium Reagents with Activated Aryl Chlorides

![Image](57x385 to 288x582)

Organolithium reagents and aryl chlorides (Scheme 1). In this case, the use of tetramethylethylene diamine (TMEDA) as an activating agent and a slightly higher temperature (35 or 40 °C) was prepared by direct metalation9 of methoxymethyl (MOM)-protected phenol (see SI for details). In addition, direct lithiation of furan led to furyllithium which was used for the reaction with chlorides 1kJ to give the corresponding cross-coupling products in good yields and high selectivity.

Table 2. Optimization of the Cross-Coupling of PhLi with Electron-Rich Aryl Chloride 1m (see also SI, Table S2)

<table>
<thead>
<tr>
<th>entry</th>
<th>[Pd]</th>
<th>t (°C)</th>
<th>reaction time (h)</th>
<th>conv (%)</th>
<th>2m:6:7</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Pd2(dba)3/L3</td>
<td>rt</td>
<td>40 min4</td>
<td>&lt;1</td>
<td>&lt;1:0:0</td>
</tr>
<tr>
<td>2</td>
<td>Pd-PEPPSI-IPent</td>
<td>rt</td>
<td>40 min4</td>
<td>61</td>
<td>57:4:0</td>
</tr>
<tr>
<td>3</td>
<td>Pd2(dba)3/L3</td>
<td>40</td>
<td>3.5 h5</td>
<td>full</td>
<td>93:1:6</td>
</tr>
<tr>
<td>4</td>
<td>Pd-PEPPSI-IPent</td>
<td>35</td>
<td>3.5 h5</td>
<td>full</td>
<td>92:4:4</td>
</tr>
</tbody>
</table>

1-Chloronaphthalene (1b) was selectively coupled with phenyllithium in high yield (91–98%), with no formation of the regioisomer 2a. Notably, when this reaction is performed on gram scale (10 mmol, 1.63 g) employing 2 mol % of catalyst B, product 2b is still obtained, after 6 h at rt, with similar selectivity and yield. The electron deficient 1-chloro-4-(trifluoromethyl)benzene 1c also underwent clean coupling with phenyllithium giving high isolated yields (catalyst A: 89% and catalyst B: 95%) of the trifluoromethylated biaryl scaffold 2c. Heteroaryl lithium compounds were also successful coupling partners. Thus, commercially available thiophenyllithium was shown to allow the cross-coupling with 1d, 1e, 1f giving rise to compounds 2d, 2e, and 2f with high yields and excellent selectivity (Scheme 1). In this case, the use of tetramethylethylenediamine (TMEDA) as an activating agent and a slightly higher temperature (40 °C) was necessary due to the reduced reactivity of this organolithium reagent.11 Highly gratifying were the results observed with sterically hindered ortho-substituted organolithium reagents as in the preparation of compounds 2g, 2h, 2i, and 2j where in all cases the cross-coupling reaction proceeded at rt, in high yield and without loss of selectivity. It should be emphasized that this new methodology is compatible with currently available procedures to access organolithium species. For example, aryl lithium reagents used in the preparation of biaryls 2g–2i were prepared via halogen–metal exchange, and the one used in the synthesis of 2j was prepared by direct metalation9 of methoxymethyl (MOM)-protected phenol (see SI for details). In addition, direct lithiation of furan led to furyllithium which was used for the reaction with chlorides 1kJ to give the corresponding cross-coupling products in good yields and high selectivity.

To explore the effectiveness of this cross-coupling with respect to the reactivity of the aryl chloride, we examined a more electron-rich aryl chloride such as 1-butyl-4-chlorobenzene 1m (Table 2).

(17) For illustrative examples on the substrate scope limitations, see Supporting Information, Scheme S1.
efficiently at this temperature, the difference could be exploited for selective cross-coupling with different chlorides present in the same molecule.

A range of deactivated aryl chlorides could be coupled with different organolithium reagents using these optimized conditions (Scheme 2).

Scheme 2. Pd-Catalyzed Cross-Coupling of Aryl Lithium Reagents with Deactivated Aryl Chlorides

Sterically hindered ortho-substituted aryl chloride 1n reacted in high selectivity with phenyllithium providing biaryl 2n in good yield using both optimized catalytic systems. The cross-coupling of PhLi with 4-methoxychlorobenzene 1o, a reluctant chloride coupling partner, using Pd2(dba)3/Xphos gave rise to a large amount of homocoupling side product. Nonetheless, we could reduce the formation of this side product to <2% by the use of Pd-PEPPSI-IPent as a catalyst which afforded product 2o in excellent yield (94%). Similarly 1-chloro-3-methoxybenzene 1p was readily arylated in good yield employing catalytic system A. Twofold arylation of 1,4-dichloro-2,5-dimethylbenzene, to form 2q, was achieved in high selectivity and good yield exemplifying the applications of this catalytic system in multiple couplings. Triaryl compound 2q is a precursor for a variety of indenofluorene derivatives with application in electronic (e.g., thin film transistor) and optoelectronic materials (e.g., organic light emitting diodes, OLED). A remarkable result is that the very hindered bis-ortho-substituted biaryl 2r was readily formed under mild conditions and high yield by reaction between aryl chloride 1r and 2-methoxyphenyllithium, employing catalytic system A. Finally, highly selective and fast coupling was also obtained in the reaction between furyllithium and 4-methoxy-chlorobenzene using the same catalytic system.

In our previous work, we described that the cross-coupling of 1-bromo-4-chlorobenzene 8 with PhLi using the Pd2(dba)3/P(tBu)3 catalyst system proceeds selectively, leaving the chloride untouched. To demonstrate the complementarity of our new methodology, we further converted chloride 9, in a subsequent cross-coupling, into heterotriaryl 10 in high yield by reaction with 2-furyllithium in the presence of the Pd-PEPPSI-IPent catalyst (Scheme 3).

In summary, we have shown the direct cross-coupling of aryl-lithium reagents with aryl chlorides in high yields and excellent selectivity. The methodology is based on the use of commercially available catalytic systems, Pd2(dba)3/XPhos or air stable Pd-PEPPSI-IPent. The reactions take place under mild conditions and feature short reaction times. The low cost and availability of both aryllithium reagents and aryl chlorides, together with the selectivity of this new methodology, make it a valuable alternative for well-established cross-coupling procedures.

Acknowledgment. The Netherlands Organization for Scientific Research (NWO-CW), National Research School Catalysis (NRSC-C), European Research Council (ERC Advanced Grant 227897 to B.L.F.), Royal Netherlands Academy of Arts and Sciences (KNAW), and Ministry of Education Culture and Science (Gravity program 024.601035) are acknowledged for financial support. C.V. was supported by an Intra-European Marie Curie fellowship (FP7-PEOPLE-2011-IEF) (Contract No. 300826).

Supporting Information Available. Optimization tables, experimental procedures, and characterization data. This material is available free of charge via the Internet at http://pubs.acs.org.

The authors declare no competing financial interest.