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Bioconjugation of Supramolecular Metallacages to Integrin Ligands for Targeted Delivery of Cisplatin

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Abstract

Cisplatin occupies a crucial role in the treatment of various malignant tumours. However, its efficacy and applicability are heavily restricted by severe systemic toxicities and drug resistance. Our study exploits the active targeting of supramolecular metallacages to enhance the activity of cisplatin in cancer cells while reducing its toxicity. Thus, Pd₂L₄ cages (L = ligand) have been conjugated to four integrin ligands with different binding affinity and selectivity. Cage formation and encapsulation of cisplatin was proven by NMR spectroscopy. Upon encapsulation, cisplatin showed increased cytotoxicity *in vitro*, in melanoma A375 cells overexpressing αvβ3 integrins. Moreover, *ex vivo* studies in tissue slices indicated reduced toxicity towards healthy liver and kidney tissues for cage-encapsulated cisplatin. Analysis of metal content by ICP-MS demonstrated that encapsulated drug is less accumulated in these organs compared to the 'free' one.

Introduction

Recently, supramolecular coordination complexes (SCCs) have attracted exploration in the fields of drug delivery, biomolecular recognition and imaging.^{1,2} Thus, the number of reports on the bioactivity of three-dimensional SCCs with different shapes, including helicates,³ metallacages,⁴ cubes,⁵ prisms⁶ and capsules,⁷ and different compositions - (M_nL_m) where M is usually Fe(II), Pd(II), Pt(II), or half-sandwich organometallic clips based on Ru(II), Os(II), or Ir(III) and Rh(III), and L is the ligand of the coordination complex - has substantially increased. In this context, supramolecular metallacages offer several properties making them attractive candidates for future drug delivery systems.⁸ These discrete chemical entities, at variance with metal organic framework (MOFs), feature a secure cavity to host small drug molecules, while the ability to modify the ligand structure both pre- and post-self-assembly allows for fine tuning of the overall properties like solubility in aqueous environment.⁹ Despite these promising properties, the full potential of metallacages as drug delivery vehicles has not been explored so far.²

In 2012, Crowley and co-workers reported on the encapsulation properties of the anticancer drug cisplatin within Pd_2L_4 cages (L = 2,6-bis(pyridine-3-ylethynyl)pyridine as the bidentate ligand), studied by NMR and X-ray diffraction.¹⁰ Platinum-based anticancer drugs are the mainstay of chemotherapy regimens in clinic. Nevertheless, the efficacy of cisplatin is badly affected by serious systemic toxicities and drug resistance, and its pharmacokinetics is still under consideration.¹¹

Therefore, cisplatin encapsulation in metallacages could enable a better delivery approach, minimizing the systemic toxicity of the drug while reducing its speciation. More recently, we explored similar cationic $[Pd_2L_4]^{4+}$ systems and developed the *exo*-functionalization of the ligand scaffold to add different functionalities (e.g. fluorescent tags).^{12,13} Structural studies by ¹H NMR and X-ray diffraction were performed demonstrating encapsulation of cisplatin.^{14,15} Furthermore, the cytotoxicity of the $[Pd_2L_4]^{4+}$ cages have been tested *in vitro* against a small panel of human cancer cells, showing scarce or moderate antiproliferative activities depending on the ligand scaffold.¹⁵ However, in order to achieve tumour selectivity, we envisaged the conjugation of the cages to cancer-cell-specific ligands – such as used in virus-like particles¹⁶ – as the best strategy. Thus, we recently reported the first example of bioconjugation of self-assembled Pd_2L_4 cages to a linear model peptide.¹⁷

Here we extended our investigation to the bioconjugation of the cages to integrin binding ligands. Integrins have been extensively studied as drug targets, interacting with the extracellular matrix (ECM) and

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3 thereby regulating many cellular functions, such as proliferation, migration, and survival.^{18,19} In mammals,
4 there are eight RGD recognizing integrin subtypes among them $\alpha\beta3$ and $\alpha5\beta1$, expressed in cancer tissue
5 and promoting tumour metastasis.¹⁹ Over the years, numerous small peptidic or peptidomimetic molecules
6 were designed and used to selectively address various integrin sub-types to treat different pathologies or
7 for imaging.²⁰⁻²⁶ As an example, the cyclic pentapeptide *cyclo*(RGDFV)²⁷ and the related Cilengitide²⁸ were
8 found to be potent ligands for the integrins $\alpha\beta3$, $\alpha\beta5$, and $\alpha5\beta1$.^{25,29}

14 It is worth mentioning that direct tethering of integrin binding peptides to cytotoxic platinum
15 complexes have been also reported. For example, *Lippard et al.* explored the targeting ability of linear and
16 cyclic RGD and Asn–Gly–Asp (NGR) integrin targeting ligands, and synthesized a Pt(IV) series comprising
17 mono- and difunctionalized complexes, which improved the cisplatin anticancer activity, once the Pt(IV)
18 compound is reduced to cisplatin in the intracellular environment.³⁰ More recently, other Pt(II) and Pt(IV)
19 complexes, functionalized with cyclic and multimeric RGD-containing peptides, have been described.^{31,32}
20 Interestingly, conjugation of a photoactivatable Pt(IV) prodrug to a cyclic RGD-containing peptide, led to
21 increased phototoxicity in melanoma cancer cells overexpressing the $\alpha\beta3$ -integrin receptor.³² However,
22 these targeting approaches do not address the problem of metallodrug *speciation* until the target is reached,
23 and, therefore, possibilities of cisplatin encapsulation into a targeted drug delivery vector are extremely
24 appealing.

26 Hence, we conjugated a Pd₂L₄ metallacage to four integrin ligands for subtype specific targeting of
27 integrin $\alpha\beta3$ (compounds **1-3**)^{26,27,33} or $\alpha5\beta1$ (compound **4**),³⁴ respectively (Figure 1).^{23,24,35,36} Cage
28 formation and cisplatin encapsulation were studied by NMR spectroscopy. The conjugated cages were first
29 studied for their integrin recognition properties using an ELISA assay. Afterwards, the three cages targeted
30 to $\alpha\beta3$ and encapsulating cisplatin were tested for their antiproliferative effects against two human cancer
31 cell lines with different levels of integrin expression, namely the melanoma cell line A375 overexpressing
32 $\alpha\beta3$, and the lung cancer cell line A549, deprived of this integrin. Moreover, the toxicity of a representative
33 cage and of its encapsulated cisplatin was evaluated *ex vivo* in healthy tissues. Finally, the accumulation of
34 cisplatin was also studied in tissues by Inductively Coupled Plasma Mass Spectrometry (ICP MS).

55 Results and Discussion

Initially, the ligand **L0** featuring a carboxylic acid group in *exo*-position has been synthesised adapting a previously reported procedure,¹² to increase the overall yield and including a protecting step of the carboxylic moiety (See Experimental for details). Thus, **L0** was used for coupling to the amino-functionalized integrin ligands **1-4** to form the cage precursors **L1-L4** (Fig. 1). The products were analysed by various methods, including NMR spectroscopy (Figures S1-S7, S10-S11, S13-S14, S16-17, S19-S20, SI). Afterwards, the cages **C0-C4** were formed by self-assembly upon addition of 2 eq. of the Pd²⁺ precursor (Pd(MeCN)₄(BF₄)₂), as proven by ¹H NMR (Fig. S8-S9 and S12, S15, S18, S21, SI), and remained stable in solution for at least 24 h.

In Figure 2, the stepwise addition of Pd²⁺ precursor to ligand **L4** enabled following the self-assembly process leading to cage **C4** formation by ¹H NMR spectroscopy. In details, to the free bioconjugated ligand **L4** (4.00 eq. Figure 2A), Pd(MeCN)₄(BF₄)₂ was added in four portions (0.50 eq., Figure 2B-E). The adjacent protons of the metal binding site undergo a significant shift of about 1 ppm (a to a' and b to b') and also the two remaining protons of the pyridyl residue (e to e' and f to f') slightly shift after binding of the Pd²⁺ ions. The more distant protons (c and d, and the protons of the integrin ligand) do not show any remarkable shift with binding of the Pd²⁺ precursor to form the cage **C4** (Figure 2E). The full consumption of the free ligand after an addition of 2.00 eq. Pd(MeCN)₄(BF₄)₂ clearly indicates a ratio of ligand to Pd of 2:1, which could be explained by the formation of cage **C4** (Pd₂L₄).

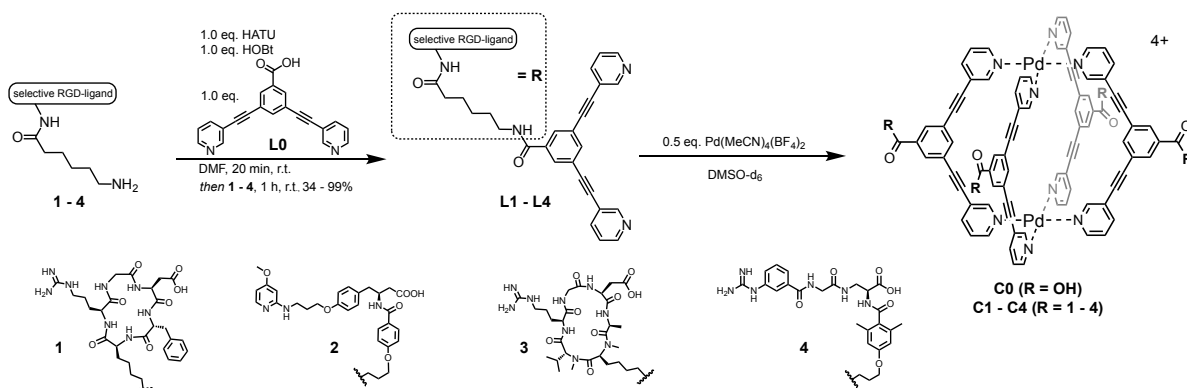


Figure 1. Synthesis of the bioconjugated ligands **L1-L4** and of cages **C0** and **C1-C4** via metal-mediated self-assembly. (HATU = hexafluorophosphate azabenzotriazole tetramethyl uronium, HOBt = hydroxybenzotriazole). Compounds **1-4** are the integrin binding ligands for $\alpha v\beta 3$ (**1-3**) and $\alpha 5\beta 1$ (**4**). The integrin ligands were originally described in the following literature: **1**^{27,35}, **2**^{24,33}, **3**²⁶, and **4**^{23,34}.

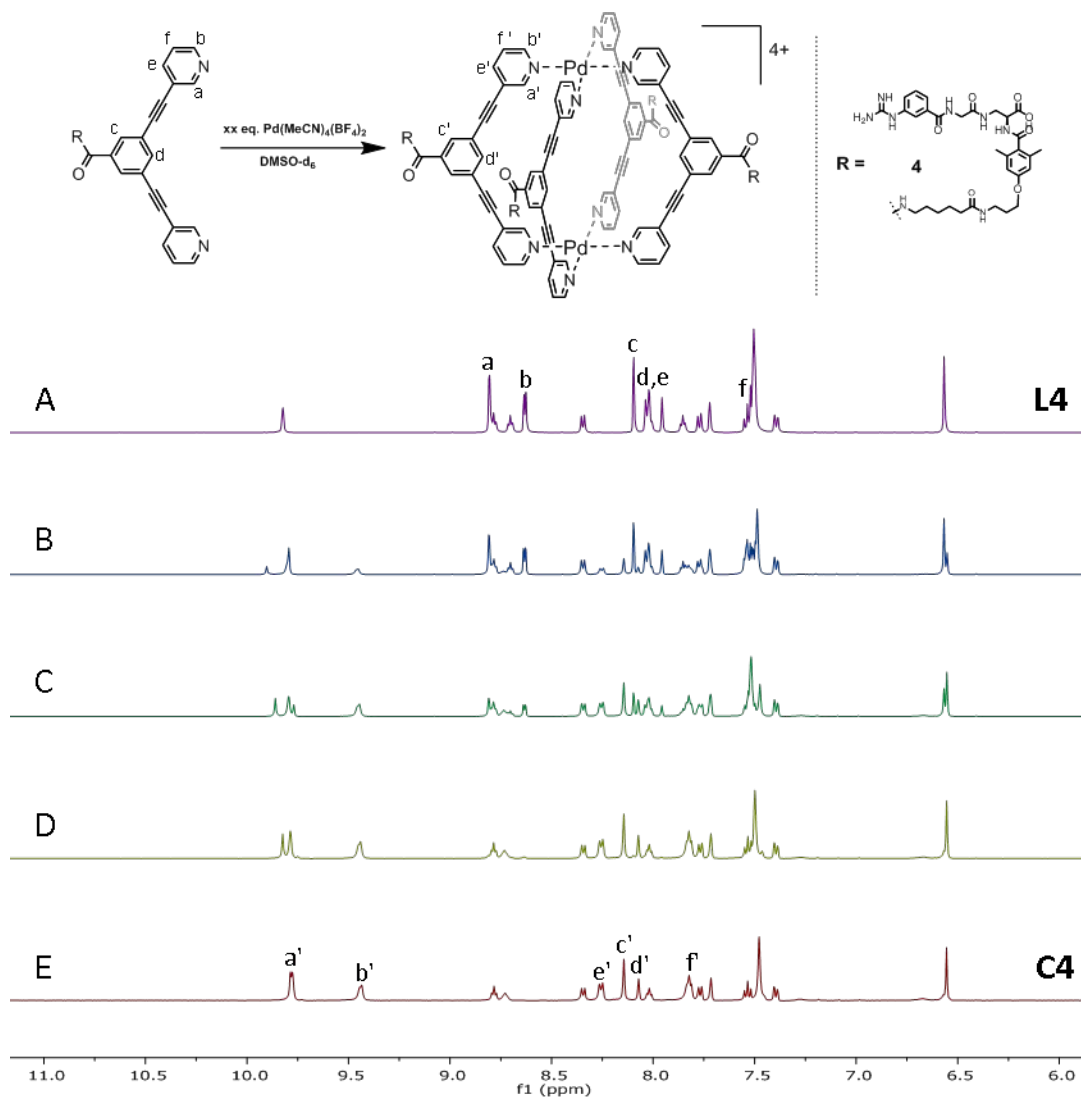


Figure 2. Titration of ligand **L4** with $\text{Pd}(\text{MeCN})_4(\text{BF}_4)_2$ and self-assembly process leading to formation of cage **C4** followed by ^1H NMR spectroscopy. A. Spectrum of the bioconjugated metal ligand **L4** (4.00 eq.) in DMSO-d_6 . Addition of the $\text{Pd}(\text{MeCN})_4(\text{BF}_4)_2$ (B. 0.50 eq.; C. 1.00 eq.; D. 1.50 eq.; E. 2.00 eq.; as a solution in DMSO-d_6) results in the step-wise self-assembly of cage **C4**.

The bioconjugated cages **C1-C4** as well as their precursors **L1-L4** were assessed for their integrin binding affinity and selectivity by an established ELISA-like solid-phase binding assay.^{25,37} Data are reported in Table 1 and show that **C1-C3** $\alpha\beta 3$ and **C4** $\alpha 5\beta 1$ complexes feature strong binding, as expected from the original binding affinity of the parent ligands **L1-L4**,²⁵ while the unsubstituted cage **C0** exhibited no binding affinity at all. Compared to Cilengitide,²⁸ the metallacages **C1-C4** show higher selectivity for their target integrin with affinities in the low nanomolar range. The observation that multimerization is improving the

binding affinity of ligands is well-known and often investigated,^{38,39} this was also found by us for RGD-containing ligands very early.^{40,41} The role of multimerization is especially important for integrins⁴² as multimeric integrin ligands might help to enter cells by endocytosis as it is used by viruses such as foot and mouth disease⁴³ and used for virul-like-particles.¹⁶ Within the group of $\alpha\nu\beta 3$ active cages **C1-C3**, the nature of the integrin binding ligand has an influence on the discriminating selectivity towards the other integrins. Whereas the cyclic peptides in **C1** and **C3** show increased affinity for all integrins upon cage formation, the peptidomimetic **C2** displays in addition an enhanced selectivity for its target integrin $\alpha\nu\beta 3$. The peptidomimetic $\alpha 5\beta 1$ targeting cage **C4**, as well as the precursor **L4**, show very high selectivity for the integrin $\alpha 5\beta 1$. The activity of **L4** for $\alpha 5\beta 1$ was far below 1 nM and might additionally be attributed to any unspecific interaction between **L4** and the integrin.

Table 1. IC₅₀ values of metallacages **C0-C4** and their ligand precursors **L1-L4** for binding to RGD-recognizing integrin subtypes $\alpha 5\beta 1$, $\alpha\nu\beta 3$, $\alpha\nu\beta 5$ and $\alpha\nu\beta 6$.

compound	IC ₅₀ [nM] ^[a]			
	$\alpha 5\beta 1$	$\alpha\nu\beta 3$	$\alpha\nu\beta 5$	$\alpha\nu\beta 6$
C0	>10000	>10000	>10000	>10000
C1	20±2	2.1±0.8	49 ± 6	159±18
C2	1025±331	0.98±0.24	6580 ±911	644±66
C3	193±62	0.38±0.01	24±6	13±2
C4	1.77±0.15	>10000	>10000	2370±303
L1	215±71	24±5	147±9	479±39
L2	66±12	3.3 ±1.3	827±118	148±11
L3	396±127	0.87±0.31	156±20	64±13
L4	<1	>1000	>10000	1361±151
Cilengitide ^[b]	14.9±3.1	0.61±0.06	8.4±2.1	2050±640

^[a] The reported IC₅₀ values were calculated using a sigmoidal fit to 16 data points, obtained from two serial dilution rows, by using Origin software package. All IC₅₀ values determined were referenced to the affinity of the internal standard and are presented as mean ± SD. ^[b] IC₅₀ values of Cilengitide as previously reported.²⁵

Previously reported results on unsubstituted Pd₂L₄ cages showed that they are able to encapsulate two molecules of cisplatin as determined by X-ray diffraction and NMR spectroscopy.^{10,15} Thus, we also studied the cisplatin encapsulation properties of the protected cage **Bz-C0** by ¹H NMR (Fig. S22) and ¹H DOSY NMR (Fig. S23-S27). The use of the protected cage instead of cage **C0** is due to its higher solubility

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3 in the selected NMR conditions. In the ^1H NMR spectrum in Fig. S22, the inward directed pyridyl proton
4 signals undergo shifts upon encapsulation of 2 eq. of cisplatin, which are consistent with previously reported
5 results.¹⁵ Specifically, H_a undergoes a downfield shift of $\Delta\delta = 0.0048$ ppm, and H_b of $\Delta\delta = 0.0148$ ppm.
6
7 Instead, for the benzyl proton H_c an upfield shift of $\Delta\delta = 0.114$ ppm was observed (Fig. S22). Furthermore,
8
9 the titration of cisplatin (up to 2 eq.) to a cage solution shows marked differences in the ^1H DOSY NMR
10 spectra of free cisplatin, due to its encapsulation (Fig. S23-S27). Further addition of cisplatin (3 eq.) shows
11
12 reappearance of the free cisplatin signals (Fig. S27). Formation of the host-guest complex was also
13
14 supported by ^{195}Pt NMR spectroscopy where the disappearance of the classical Pt resonance of free
15
16 cisplatin at -2080 ppm was observed upon addition of cage **Bz-C0** (Figure S28).
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20 To validate our targeting concept, the antiproliferative effects of the coordination cages **C1-C3**,
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22 alone and encapsulating cisplatin [cisplatin \subset (**C1-3**)] were evaluated against two different cell lines with
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24 different $\alpha\text{v}\beta 3$ integrin expression pattern, in comparison to free cisplatin (see SI for experimental details).
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26 Specifically, the malignant melanoma cell line A375 has been shown to overexpress $\alpha\text{v}\beta 3$,⁴⁴ while the
27
28 human lung cancer cell line A549 is reported not to express it.⁴⁵ The latter cell type is also resistant to
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30 cisplatin. Thus, cells were treated with increasing concentrations of cages and ligands; concentrations
31
32 higher than 60 μM were usually not reached due to the toxic effect induced by the presence of DMSO,
33
34 particularly in the A375 cells. Drug encapsulation in each cage system was assessed by ^1H NMR prior each
35
36 experiment and confirmed encapsulation of 2 eq. of cisplatin per eq. of cage (data not shown). It should be
37
38 mentioned that cisplatin stock solutions were freshly prepared in aqueous solution (1 mM) to avoid ligand
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40 exchange reactions with DMSO.
41

42 The results are summarized in Table 2 and show that cages **C1-C3** alone are scarcely toxic to the
43
44 two cell lines, while cisplatin is moderately toxic only in the A375 cells ($\text{EC}_{50} = 33.9 \pm 2.9 \mu\text{M}$). Moreover,
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46 while treatment with cisplatin encapsulated in the “untargeted” cage **C0** ([cisplatin \subset (**C0**))] does not change
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48 the drug’s antiproliferative effects, cisplatin encapsulated in the bioconjugated cage **C2** significantly
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50 increases its cytotoxic potency against the $\alpha\text{v}\beta 3$ integrin expressing A375 cells (ca. 2-fold more potent with
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52 respect to free cisplatin). Instead, the same [cisplatin \subset (**C2**)] formulation does not increase the cisplatin
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54 toxicity against the A549 cells with no expression of $\alpha\text{v}\beta 3$. Concerning cages **C1** and **C3**, none of them
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56 increase significantly the cytotoxicity of cisplatin in the A375 cell line, which could be related to the lower
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58 selectivity against $\alpha 5\beta 1$ and/or $\alpha\text{v}\beta 5$ (Table 1). This result could also be due to differences in stability and
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3 drug release properties of the cages in the biological environment. Finally, it is worth mentioning that the
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5 Pd²⁺ precursor, as well as the different ligands **L0** and **L1-L4** were also studied and resulted to be non-toxic
6
7 at the highest tested concentrations in both cell lines. Representative values are reported in Table 2 for **L0**.
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11 **Table 2.** Antiproliferative activity (EC₅₀ values) of different cages and cage:cisplatin (1:2) formulations in
12 human A375 and A549 cancer cells compared to cisplatin after 24 h incubation.
13

Treatment	EC ₅₀ [μM] ^[a]	
	A375 (melanoma) αvβ3 ++	A549 (lung) αvβ3 -
Ligand L0	> 50	> 100
cage C0	> 50	82.6 ± 15.1 ^[b]
[cisplatin⊂(C0)]	31.5 ± 3.0	> 60
cage C1	> 50	> 60
[cisplatin⊂(C1)]	32.4 ± 1.2	> 60
cage C2	> 50	> 60
[cisplatin⊂(C2)]	15.80 ± 3.35*	> 60
cage C3	> 50	> 60
[cisplatin⊂(C3)]	29.5 ± 0.6	> 60
cisplatin	33.9 ± 2.9	63.2 ± 5.2

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33 ^[a] The reported EC₅₀ values are presented as mean (± SEM) of at least four independent experiments.

34 ^[b] Value taken from reference [12]. * (p < 0.01) indicate the difference is significant when compared to samples treated
35 with cisplatin only.
36

37
38 In order to constitute an optimal drug delivery system, metallacages should be deprived of intrinsic
39 toxic effects in normal tissues, and preferably reduce the toxicity exerted by the encapsulated drug. Thus,
40 as proof of concept, cage **C0**, alone or encapsulating cisplatin, was tested for its toxicity on healthy rat
41 kidney and liver tissues *ex vivo* using the precision-cut tissue slices (PCTS) technique.⁴⁶⁻⁵⁰ PCTS are thin
42 (150-250 μm) slices of viable tissue. In this model, all cells remain in their natural environment maintaining
43 the original cell-cell and cell-matrix contacts, which are absent in classical 2D cell cultures *in vitro*. This
44 method is a FDA-approved model for drug toxicity and metabolism studies, and is also useful to determine
45 drug uptake/efflux mechanisms.⁴⁶⁻⁵⁰ Recently, we have successfully used the PCTS method to study the
46 toxic effects of experimental anticancer organometallic compounds, including aminoferrocene-containing
47 prodrugs, ruthenium-based kinase inhibitors, as well as gold(I) carbenes and gold(III) cyclometalated
48 cytotoxic agents.⁴⁶⁻⁵⁰
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Thus, the viability of the liver and kidney slices was determined by incubating the PCTS with different concentrations of cage **C0**, alone or encapsulating cisplatin (cage:cisplatin = 1:2), for 24 h and measuring the ATP/protein content (see SI for details). Cisplatin was also administered in the same range of concentrations for comparison. The obtained results for liver and kidney are presented in Fig. 3 (A and B). As expected, and in line with previous results,⁴⁹ cisplatin alone causes a marked reduction of slice viability in both organs. Instead, cage **C0** was substantially not toxic even at the highest tested concentration in the liver slices. Ligand **L0** and the Pd²⁺ precursor were also non-toxic above 100 μM (data not shown). Of note, encapsulated cisplatin showed a significantly reduced toxicity in both liver and kidney slices compared to free cisplatin.

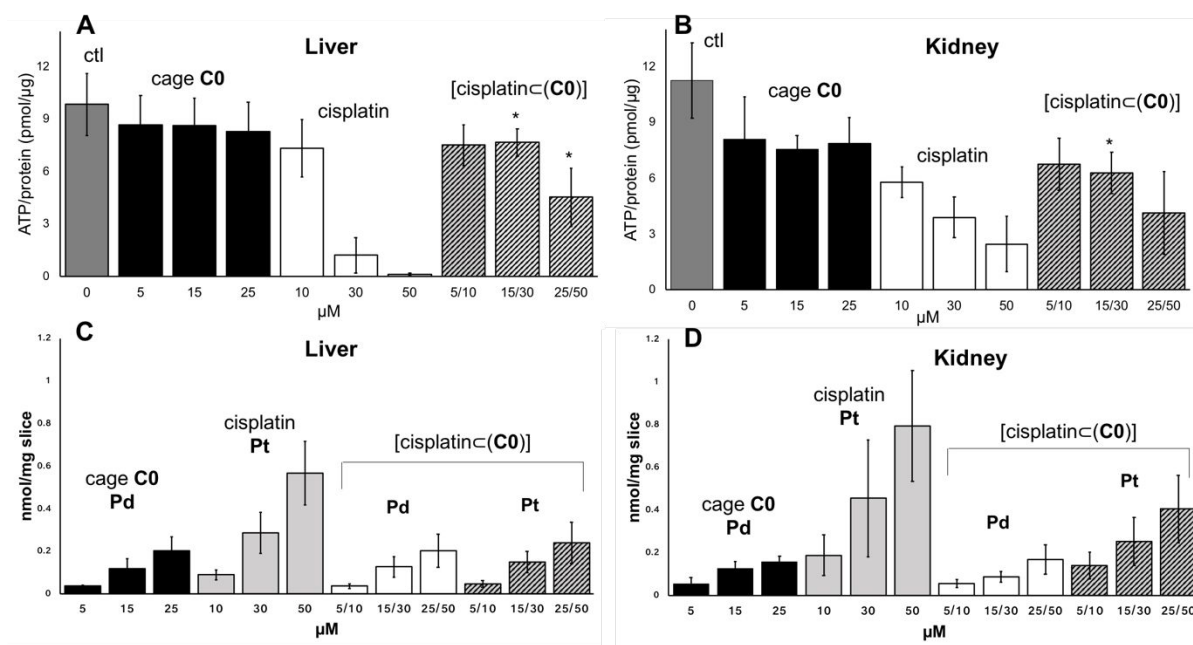


Figure 3 – A and B: Viability of Precision Cut Tissue Slices (PCTS) from liver (A) and kidney (B), treated for 24 h with different concentrations of cisplatin (5, 10 and 25 μM), cage **C0** (10, 30 and 50 μM) and encapsulated cisplatin [cisplatin<(C0)] ([cage]/[cisplatin] = 5/10, 15/30, 25/50 μM). **C and D:** Total metal content (Pd and Pt) determined by ICP-MS in PCTS of liver (C) and kidney (D) treated for 24 h with different concentrations of cisplatin, cage **C0** and encapsulated cisplatin ([cisplatin<(C0)]). The error bars show the standard deviation of at least three independent experiments. For statistical analysis the Two Independent Sample t-Test was applied. * ($p \leq 0.01$) indicates the difference is significant when compared to its control (treatment with cisplatin at the same concentration).

In order to evaluate the relation between toxicity and intracellular metal accumulation, we determined the Pt and Pd content of tissue slices exposed to cisplatin alone or encapsulated in cage **C0** by ICP-MS. Thus, PCTS were incubated for 24 h in the same conditions as for the ATP determination described

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3 above. In PCTS, the metal concentration increases in a concentration dependent manner (Fig. 3, C and D).
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5 Overall, as expected,⁵¹ the Pt content is higher in kidney (Fig. 3D) than in liver (Fig. 3C). However, the
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7 accumulation of Pd is somewhat higher in the liver than in the kidney PCTS. Moreover, the results suggest
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9 that Pd accumulation is not influenced by cisplatin encapsulation. As it can be observed in Fig. 3C, in the
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11 case of liver slices, cisplatin encapsulated in cage **C0** is markedly less accumulated than free cisplatin.
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13 Similar conclusions can be made for kidney samples although the differences are not as significant (Fig.
14
15 3D). Thus, the obtained results are another indirect evidence that cisplatin encapsulation in the metallacage
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17 is mostly preserved even in biological environment.

18
19 In the liver, the content of Pd and Pt in the [cisplatin-(**C0**)] treated slices is roughly in accordance with the
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21 predicted metal stoichiometry (Pd:Pt). However, in the kidney, the Pt content is higher than the Pd content,
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23 and this result should be further investigated. Possibly, Pd is partly excreted from the tissue over 24 h, while
24
25 Pt is not (or less). Alternatively, a part of cisplatin may be released from the cage before uptake and enter
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27 the tissue more efficiently than in the liver. In this latter case the overall Pt content may be the result of
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29 accumulation of both encapsulated and free cisplatin. However, this latter explanation may be less likely in
30
31 liver as such an effect is not seen in this organ.

32
33
34 In conclusion, we herein present an integrin selective supramolecular drug delivery system for cisplatin
35
36 based on the Pd₂L₄ metallacage scaffold. Encapsulation of cisplatin in the RGD-modified metallacage
37
38 exhibits increased anticancer effects *in vitro*, and the metallacaged cisplatin itself has a decreased toxicity
39
40 against healthy tissues *ex vivo*. The latter observation is explained on the basis of reduced platinum uptake
41
42 in healthy tissues. Although further studies are necessary to fully characterize the properties of metallacages
43
44 as drug delivery systems, including their stability in physiological conditions and drug release properties,
45
46 toxicity of cages conjugated with integrin ligands in healthy tissue, as well as study of the effects in *in vivo*
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48 models, our work constitutes a first proof-of-concept of the possible use of supramolecular coordination
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50 complexes for directed drug delivery.
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Experimental

Synthesis

Synthesis and analysis of ligand L0

Ligand **L0** has been synthesised adapting a previously reported procedure,¹² to increase the overall yield and including a protecting step of the carboxylic moiety (Scheme S1). The starting compound benzyl 3,5-dibromobenzoate was synthesised according to a previously reported procedure by benzyl protection of 3,5-dibromobenzoic acid,⁵² and fully characterized by NMR (see supplementary information for details and Figures S1-S6).

Synthesis of ligands 1-4

In this section we report the general procedures (GPs) for the synthesis of the integrin binding ligands. For details and analysis, we refer to the supplementary material.

GP1 Loading of TCP-resin. Peptide synthesis was carried out using 2-chlorotrityl chloride resin (2-CTC-resin) (0.9 mmol/g) following standard Fmoc-strategy. Fmoc-Xaa-OH (1.20 eq.) were attached to the resin with DIPEA (2.50 eq.) in anhydrous CH₂Cl₂ (10 mL/g resin) for 2 h at r.t. Remaining 2-chlorotrityl chloride groups were capped by adding a mixture of MeOH, DIPEA (5:1; v/v; 1.0 mL/g resin) for 15 min. The resin was then filtered and washed with CH₂Cl₂ (5x) and *N*-methylpyrrolidone (NMP; 3x). The loading capacity was estimated to be 0.9 mmol/g (100 %).

GP2 On-Resin Fmoc Deprotection. The resin-bound Fmoc peptide was treated with 20% (v/v) piperidine in NMP twice (1x10 min, 1x5 min) and subsequently washed with NMP (5x).

GP3 Standard Amino Acid Coupling. A solution of Fmoc-Xaa-OH (2.00 eq.), *O*-(7-azabenzotriazol-1-yl)-*N,N,N',N'*-tetramethyluronium-hexafluorophosphate (HATU) (2.00 eq.), 1-hydroxybenzotriazole (HOBt; 2.00 eq.), and DIPEA (5.00 eq.) in NMP (10 mL/g resin) was added to the resin-bound free amine peptide and shaken for 60 min at r.t. The resin was then washed with NMP (5x).

GP4 Cleavage of Linear Peptides from the Resin. For complete cleavage from the resin, the peptides were treated with a solution of CH₂Cl₂ and hexafluoroisopropanol (HFIP; 4:1; v/v) for 15 min at r.t. (2x). The solvent was then evaporated under reduced pressure.

GP5 Peptide cyclization. To a solution of linear peptide in DMF (2 mM peptide concentration) and NaHCO₃ (5.00 eq.) diphenylphosphoryl azide (DPPA; 3.00 eq.) was added at r.t. and stirred until completion of

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3 cyclization was observed (monitored by HESI-MS, usually over night). The solvent was evaporated *in vacuo*
4 and the crude material directly used for the next step.

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6 *GP6 Cbz-deprotection in solution.* The Cbz-protected starting material was dissolved in MeOH (0.15 m), Pd
7 on carbon (0.02 eq. 10 wt.%, dry) was added and hydrogen bubbled through the reaction mixture. Reaction
8 progress was monitored by HESI-MS-analysis. After complete deprotection (usually 1-3 h) the reaction
9 mixture was filtrated over Celite and the solid residue was washed with MeOH. The solvent was removed
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14 *in vacuo.*

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16 *GP7 Coupling in Solution.* The free amine was dissolved in DMF (~0.1 mol/L) and the acid (1.20 eq.), HATU
17 (1.20 eq.), HOBT (1.20 eq.) and DIPEA (3.00 eq.) were added and the solution stirred over night or until no
18 residual free amine could be observed (monitored by HESI-MS). After removal of the solvent *in vacuo*, the
19 residue was taken up in EtOAc, washed with HCl_{aq.} (1 M, 2x) and brine. The organic phase was dried over
20 Na₂SO₄, filtrated and the solvent removed *in vacuo*. The isolated products were purified *via* RP-HPLC or
21 used as crude starting material for the next reaction step.

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27 *GP8 Removal of acid-labile side-chain protecting groups:* Cyclized peptides were stirred in a solution of
28 TFA, CH₂Cl₂, water and TIPS (80-50:15-45:2.5:2.5; v/v/v) at r.t. until complete deprotection (approx. 1 h,
29 monitored by HESI-MS). The reaction was quenched by addition of toluene and the solvent mixture was
30 evaporated under reduced pressure (2x).

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35 *GP9 On-resin N-methylation.* The Fmoc-protected peptide was washed with CH₂Cl₂ (3x), a solution of 2-
36 nitrobenzenesulfonylchloride (4.00 eq.) and 2,4,6-collidine (10.0 eq.) in CH₂Cl₂ was added and mixed for 20
37 min at r.t. The resin was washed with CH₂Cl₂ (3x) and THF *abs.* (5x). A solution containing PPh₃ (5.00 eq.)
38 and MeOH *abs.* (10.0 eq.) in THF *abs.* was added to the resin. DIAD (5.00 eq.) in a small amount THF *abs.*
39 was added stepwise to the resin and mixed for 15 min at r.t. The resin was subsequently washed with THF
40 (5x) and NMP (5x). For *o*-Ns deprotection, the resin-bound *o*-Ns-peptides were stirred in a solution of
41 mercaptoethanol (10.0 eq.) and DBU (5.00 eq.) in NMP (10 mL/g resin) for 5 min at r.t. The deprotection
42 procedure was repeated once and the resin was subsequently washed with NMP (5x).

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51 *GP10 Dde-deprotection in solution.* The orthogonal deprotection of Dde was performed using a solution of
52 hydrazine hydrate (2 vol.-%) in DMF for 30 min at r.t. The progress of the reaction was monitored by HESI-
53 MS. After completion of the reaction, the solvent was removed under reduced pressure.

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3 *GP11 Boc-, ^tBu-deprotection in solution.* The starting material was dissolved in CH₂Cl₂ (~1 mL/100 mg) and
4 an equal volume of TFA added at r.t. The solution was stirred for 30-60 min. The reaction was quenched by
5 the addition of toluene and the solvent was evaporated under reduced pressure. The latter step was
6 repeated once.
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10 11 12 Synthesis of ligands **L1-L4**

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14 3,5-bis(pyridin-3-ylethynyl)benzoic acid (1.00 eq.), HATU (*O*-(7-azabenzotriazol-1-yl)-*N,N,N',N'*-
15 tetramethyluronium-hexafluorophosphate, 1.00 eq.) and HOBt (1-hydroxybenzotriazole, 1.00 eq.) were
16 dissolved in DMF (50 mM, referred to 1.00 eq.) and DIPEA (*N,N*-diisopropylethylamine, 2.00 eq.) was
17 added. The solution was stirred at room temperature for 20 min and subsequently added to a solution of the
18 corresponding amine (50 mM in DMF, 1.00 eq.) in DMF. The reaction mixture was stirred for 1 h at r.t., the
19 solvent was removed *in vacuo* and the crude product was purified *via* semi-preparative RP-HPLC. For
20 details and analysis, we refer to the supplementary material.
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30 Synthesis of cage **C0**

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32 A solution of [Pd(NCCH₃)₄](BF₄)₂ (22.2 mg, 0.05 mmol, 2.00 equiv.) and ligand **L0** (32.4 mg, 0.10 mmol,
33 4.00 equiv.) in DMSO was stirred for 1 h at room temperature (Scheme S3). Following precipitation by
34 addition of acetone and diethyl ether, the solid was then filtrated and washed with diethyl ether to yield the
35 cage compound as an off-white solid (42.0 mg, 23.0 μmol, 92%). The product was characterized by NMR
36 spectroscopy and high-resolution mass spectrometry (Figures S7-S9).
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43 Synthesis of cages **C1-C4**

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45 Synthesis of cages **C1-C4** was achieved by self-assembly adapting the procedure reported for cage **C0**
46 described above. The starting material (compounds **L1-L4**, ~2 μmol, 4.00 eq.) was weighed into an NMR-
47 tube, respectively, and solubilized in DMSO-*d*₆ (0.4 mL). A solution of Pd(MeCN)₄(BF₄)₂ in DMSO-*d*₆
48 (20.54 mM, 2.00 eq.) was added to the solution in the NMR-tube and homogenized. In the ¹H-NMR, the
49 starting material was completely converted to the corresponding metallacage (**C1-C4**). For the ligands **L1**,
50 **L3** and **L4** the molecular weight was calculated as one-fold TFA-salts, for ligand **L2** a calculation without
51 TFA-salt gave full consumption of the starting material. Due to their chromatographic instability, the
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3 formation procedure was optimized in NMR-solvent (DMSO-d₆) and analyzed without isolating the product.
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5 Hereafter ELISA-assays were performed directly with these solutions. ¹H-NMRs are shown in Figure 2 and
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7 Figures S10-S20 of the Supplementary Information.
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14 Titration of Pd(MeCN)₄(BF₄)₂ to a solution of ligand L4.

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16 The exclusively tetrameric cage formation of **C4** (Pd₂L₄) is proven by a titration experiment. To a solution
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18 of ligand **L4** (2.89 mg, 2.89 μmol, 4.00 eq.) in DMSO-d₆ (0.4 mL) a solution of Pd(MeCN)₄(BF₄)₂ (14.3 mM)
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20 in DMSO-d₆ was added in four portions (each 0.36 μmol, 23.8 μL). After each addition a ¹H-NMR spectrum
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22 was measured (see Figure 2).
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26 **Encapsulation studies**

27 ¹H-NMR spectroscopy

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29 The benzoate protected Pd₂L₄ cage **C0-Bz** (8 mg, 36 μmol, 1.00 eq.) was dissolved in DMF-d₇ (1mL) and
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31 the ¹H-NMR (500 MHz, 64 scans) spectrum was recorded. Cisplatin (2 mg, 72 μmol, 2.00 eq.) was added
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33 and the yellow solution was stirred for 10 min. The ¹H-NMR (500 MHz, 64 scans) spectrum of the sample
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35 was recorded and compared with the previous one (Figure S21).
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38 ¹H-DOSY NMR spectroscopy

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40 The benzoate protected ligand **L0-Bz** (15.7 mg, 37.9 μmol, 4.00 eq.) was dissolved in DMF-d₇ (2.63 mL,
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42 14.4 mM) and a solution of Pd(MeCN)₄(BF₄)₂ (8.41 mg, 18.9 μmol, 2.00 eq.) in DMF-d₇ (461 μL, 41.1 mM)
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44 was added and a ¹H-NMR (400 MHz) spectrum was measured to ensure cage-formation. For each ratio,
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46 cisplatin was exactly measured out in a glass vial and the previously prepared cage solution was added to
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48 get a final cage:cis-Pt ration of 1:1, 1:2, 1:3. The solution was mixed for about 5 min to guarantee complete
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50 dissolution of cisplatin, subsequently a ¹H-NMR and a DOSY-NMR was measured (Figure S22-S26).
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53 **Integrin binding studies**

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55 The affinity and selectivity of integrin ligands were determined by a solid-phase binding assay applying a
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57 previously described protocol³⁷ that involves coated extracellular matrix (ECM) proteins and soluble
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3 integrins. The following compounds were used as internal standards: *cilengitide*²⁹, *c(f(NMe)VRGD)* ($\alpha\beta3$ –
4 0.54 nM, $\alpha\beta5$ – 8 nM, $\alpha5\beta1$ – 15.4 nM). Flat-bottomed 96-well ELISA plates (*BRAND*, Wertheim, Germany)
5 were coated overnight at 4 °C with ECM protein (1) (100 μL per well) in carbonate buffer (15 mM Na_2CO_3 ,
6 35 mM NaHCO_3 , pH 9.6). Afterwards, each well was washed with PBS-T buffer (phosphate-buffered
7 saline/Tween 20, 137 mM NaCl, 2.7 mM KCl, 10 mM Na_2HPO_4 , 2 mM KH_2PO_4 , 0.01% Tween 20, pH 7.4;
8 3 \times 200 μL) and blocked for 1 h at room temperature with TS-B buffer (Tris-saline/bovine serum albumin
9 (BSA) buffer, 20 mM Tris-HCl, 150 mM NaCl, 1 mM CaCl_2 , 1 mM MgCl_2 , 1 mM MnCl_2 , pH 7.5, 1% BSA;
10 150 μL /well). Meanwhile, a dilution series of the compounds and internal standard was prepared in an extra
11 plate, ranging from 20 μM to 256 pM in 1:5 dilution steps. After washing the assay plate three times with
12 PBS-T (200 μL), 50 μL aliquots of the dilution series were transferred to each well from B-G in 6 appropriate
13 concentrations. Well A was filled with 100 μL of TSB solution (blank), and well H was filled with 50 μL of
14 TS-B buffer. Then, 50 μL of a solution of human integrin (2) in TS-B buffer was transferred to wells H–B and
15 incubated for 1 h at r.t. The plate was washed three times with PBS-T buffer, and then primary antibody (3)
16 (100 μL per well) was added to the plate. After incubation for 1 h at r.t., the plate was washed three times
17 with PBS-T. Then, secondary peroxidase-conjugated antibody (4) (100 μL /well) was added to the plate and
18 incubated for 1 h at r.t. Details on the respective solutions (1-4) for each integrin ligand are provided in the
19 supplementary material. The plate was then washed three times with PBS-T, developed by the addition of
20 SeramunBlau (50 μL /well, Seramun Diagnostic GmbH, Heidesee, Germany) and incubated for approx.
21 1 min at r.t. in the dark. The reaction was stopped with 3 M H_2SO_4 (50 μL /well), and the absorbance was
22 measured at 450 nm with a plate reader (*infinite M200 Pro*, *TECAN*). The IC_{50} value of each compound
23 resulted from a sigmoidal fit of two data rows (serial dilution rows) done by *OriginPro 9.0G* statistical
24 software. All IC_{50} values were referenced to the affinity of the internal standard.
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47 **Antiproliferative assays**

48 Human malignant melanoma cancer cell line A375 and human lung cancer cell line A549 were obtained
49 from ATCC and maintained in culture as described by the provider. The cells were cultured in DMEM
50 (Dulbecco's Modified Eagle Medium, Corning), supplemented with 10% fetal bovine serum (FBS, Eu-
51 approved South American Origin, Thermo Fisher Scientific) and 1% penicillin/ streptomycin (Gibco) at 37
52 °C and 5% CO_2 .
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To monitor the self-assembly of the cages **C0-C4** from ligands **L0-L4** and to study the subsequent cisplatin encapsulation, ¹H NMR studies were conducted directly in DMSO/H₂O with a capillary of CDCl₃ (so the solution could then directly be used for further biological studies). Thus, each cage was dissolved in 1 mL of DMSO (max 20%) in H₂O (stock solution 3 mM, 1 eq.) and added to a NMR tube containing a closed capillary of CDCl₃. Only in the case of the poorly soluble cage **C0** the initial stock solution 3 mM was prepared in 100% DMSO. ¹H NMR spectra were recorded before and after addition of 2 eq. of cisplatin to assess drug encapsulation. After confirmation of the encapsulation process, the solution was then extracted from the NMR tube and further diluted for the cell viability studies (0.1 to 60 μM) in DMEM medium. The maximal tested DMSO concentration was 0.1%, except than in the case of the least soluble cage **C0** (0.5%). In this latter case, control experiments with different concentration of DMSO were conducted to exclude possible effects on cells viability. Concentrations higher than 50 μM were not tested in A375 cells to avoid toxic effects of DMSO. Cisplatin (Sigma-Aldrich) was used as reference compound (tested between 0.1 to 100 μM), and its stock dilutions (1 mM) were freshly prepared in aqueous solution prior each experiment. For evaluation of cell growth inhibition, cells were seeded in 96-well plates (Corning) at a concentration of 15000 cells/well and grown for 24 h in 200 μL complete medium. After 24 h incubation, 200 μL of the compounds' dilutions were added to each well and the cells were incubated for additional 24 h. Afterwards, the medium was removed and 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT, Fluorochem) in 10x PBS (Phosphate Buffered Saline, Corning) was added to the cells, at a final concentration of 0.3 mg/mL and incubated for 3-4 h. Then, the MTT solution was discarded and replaced with DMSO to allow the formed violet formazan crystals to dissolve. The optical density was quantified in quadruplicate at 550 nm using a multi-well plate reader. The percentage of surviving cells was calculated from the ratio of absorbance of treated to untreated cells. The EC₅₀ value was calculated, using GraphPad Prism software, as the concentration causing 50% decrease in cell viability, using a nonlinear fitting of cell viability vs [treatment], and presented as mean ± SEM of at least three independent experiments. Statistical testing was performed using a Two Sample t-Test, to compare the [cisplatin-(**C0-3**)] treated samples with its control samples (treatment with cisplatin at the same concentration). A *p*-value of ≤0.01 was considered to be significant.

Ex vivo studies

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3 Male Wistar rats (Charles River, France) of 250-300 g were housed under a 12 h dark/light cycle at constant
4 humidity and temperature. Animals were permitted *ad libitum* access to tap water and standard lab chow.
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6 All experiments were approved by the committee for care and use of laboratory animals of the University of
7 Groningen and were performed according to strict governmental and international guidelines.
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9 Livers and kidneys were harvested (from rats anesthetized with isoflurane) and immediately placed in
10 University of Wisconsin solution (UW, Dupont Critical Care, Waukegan, IL, USA, 4°C) until further use. For
11 livers, cylindrical cores of 5 mm diameter were made using a Schlagbohrer Bosch (3 603 A 26 200). The
12 kidneys, after removal of fat were cut in half lengthwise using a scalpel, and cores of 5 mm diameter were
13 cut from the kidney cortex from each half perpendicular to the cut surface using disposable Biopsy Punches
14 (KAI medical, Japan). Precision-cut liver slices (PCLS, diameter 5 mm; thickness 250 µm; weight 5 mg;) and
15 precision-cut kidney slices (PCKS, diameter 5 mm; thickness 150 µm; weight 3 mg;) were sliced with a
16 Krumdieck tissue slicer (TSE, Bad Homburg, Germany) in ice-cold Krebs buffer at pH 7.42, enriched with
17 glucose to a final concentration of 25 mM and saturated with carbogen (95% O₂/5% CO₂).⁴⁴ PCLS and
18 PCKS were immediately moved to ice-cold UW and stored on ice until the beginning of the experiment for
19 maximally 3 h. PCLS and PCKS were preincubated individually at 37 °C for 1 h in 12-well plates in 1.3 mL
20 of Williams Medium E (WME, Gibco by Life Technologies, UK) with glutamax-1 (Gibco, Invitrogen, Paisley,
21 Scotland) supplemented with 25 mM D-glucose, supplemented with 50 µg/mL gentamycin (Gibco,
22 Invitrogen, Paisley, Scotland) for PCLS and with ciprofloxacin HCl (10 µg/mL, Sigma-Aldrich, Steinheim,
23 Germany) for PCKS. In the incubator (Sanyo CO₂/O₂ Incubator, PANASONIC, Secaucus, NJ, USA), the
24 plates were under 80% O₂ and 5% CO₂ atmosphere, while being gently shaken (90 times/min).
25 Preincubation allows the slices to restore the ATP levels⁴⁴ and is helpful to remove debris and dead cells
26 before the start of the experiments. After preincubation, the slices were moved to different well plates filled
27 with 1.3 mL of prewarmed complete WME medium and different concentrations of the tested compounds
28 and incubated for 24 h. The final concentrations of tested compounds are: Cage **C0** = 5 µM, 15 µM and 25
29 µM with 0.83% DMSO v/v; cisplatin = 10 µM, 30 µM and 50 µM; [cisplatin<(C0)] = [cisplatin]:[cage] of 5/10
30 µM, 15/30 µM and 25/50 µM with 0.83% DMSO v/v. For the final concentration of [cisplatin<(C0)].
31 Experiments were performed in triplicate using at least 3 individual rat organs. For each treatment, three
32 slices were used to assess the ATP content. It is important to note that the PCLS and PCKS of each
33 liver/kidney sample were exposed to all three compounds, as well as controls, in order to limit the influence
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3 of non-biological experimental variation. This experimental design helps to limit interindividual variation
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5 between compound's classes, as well as between compounds and controls, in the data set.
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8 Evaluation of ATP content

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10 After PCTS pre-incubation, different concentrations of cisplatin and cage **C0** or of encapsulated cisplatin
11 [cisplatin-(**C0**)] were added to the wells and the slices were incubated for 24 h. Afterwards, slices were
12 collected for ATP and protein determination, by snap freezing in 1 mL of ethanol (70% v/v) containing 2 mM
13 EDTA with pH = 10.9. After thawing, the slices were homogenized using a mini-bead beater and centrifuged.
14 The supernatant was used for the ATP assay and the pellet was dissolved in 5 M NaOH for the protein
15 assay. ATP was measured using the ATP Bioluminescence Assay kit CLS II (Roche, Mannheim, Germany).
16 The ATP content was corrected by the protein amount of each slice and expressed as pmol/ μ g protein. The
17 protein content of the PCTS was determined by the Bio-Rad DC Protein Assay (Bio-Rad, Munich, Germany)
18 using bovine serum albumin (BSA, Sigma-Aldrich, Steinheim, Germany) for the calibration curve.
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30 Statistics

31 A minimum of three independent experiments were performed using slices in triplicates from each rat tissue.
32 Statistical testing was performed with Two Sample t-Test to compare the [cisplatin-(**C0**)] treated samples
33 with its control samples (treatment with cisplatin at the same concentration). A p-value of ≤ 0.01 was
34 considered to be significant.
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41 **Metal content determination by ICP MS**

42 After incubation with the different concentrations of cisplatin, cage **C0** or of the cage/cisplatin complex
43 [cisplatin-(**C0**)], PCTS were washed with ice-cold Krebs-Henseleit buffer and snap-frozen and stored at -
44 80°C until the analysis.
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51 Sample preparation and Pt and Pd determination

52 The tissue samples were digested with 100 μ L nitric acid overnight, all samples were completely dissolved.
53 100 μ L hydrochloric acid and 800 μ L milliQ were added to produce a volume of 1 mL. Prior to analysis the
54 samples were diluted 20 times with 0.65% HNO₃/0.1% HCl.
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3 The Pt and Pd contents were quantitated applying a Perkin Elmer (Waltham, MA, USA) Sciex Elan DRC-e
4 ICP-MS instrument, equipped with a Cetac ASX-110FR autosampler, a 0.2 mL min⁻¹ MicroMist U-series
5 pneumatic concentric nebulizer (Glass Expansion, West Melbourne Vic, Australia) and a PC3 cyclonic spray
6 chamber (Elemental Scientific Inc., Omaha, NE, USA). ICP-MS RF power, lens voltage and nebulizer gas
7 and flow were optimized on a daily basis and other settings were: 1 sweep/reading, 25 readings/replicate,
8 5 replicates, 50 ms dwell time. The ¹⁹⁵Pt⁺, ¹⁹⁴Pt⁺ and ¹⁰⁵Pd⁺ isotopes were monitored. Pt and Pd
9 concentrations were determined by external calibration (0-20 ppb Pt and Pd). LODs were 0.1 µg L⁻¹ for both
10 Pt and Pd, (3*SD on blank, n=10) and the spike recovery were 102% and 99% for Pt and Pd (n = 3),
11 respectively. Pt and Pd single element PlasmaCAL standards (SCP Science, Québec, Canada) were used
12 and the standards were prepared in a mixture of 0.1% HCl and 0.65% sub-boiled HNO₃ in MilliQ water. This
13 mixture was furthermore used to dilute samples after digestion and as blank solution.
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26 Statistics

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28 A minimum of three independent experiments were performed using slices with triplicates for each condition.
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30 The PCLS and PCKS were prepared from three rats and in each experiment slices were exposed in
31 triplicate. The Pt and Pd contents were expressed as nmol/mg slice and is presented as a mean (± SD) of
32 at least three independent experiments. Pt and Pd concentrations were calculated to compare the
33 [cisplatin=(**C0**)] treated samples with its control samples (treatment with cisplatin or **C0** at the same
34 concentration).
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42
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Supporting Information

The Supporting Information, including experimental details and compounds analysis, as well as results of encapsulation studies by NMR spectroscopy, is available free of charge on the ACS Publications website at DOI: XXXX.

Authors Contributions

A.C., P. H., H. K. conceptualized the project. A.C. wrote the manuscript, supervised the synthesis of the ligands and cages, and designed the *in vitro* experiments. G.M.M.G. designed the *ex vivo* experiments. H.P. and R.B. contributed to the design of the biological experiments, cage analysis and data interpretation. J.H. performed the *ex vivo* experiments and set-up the first protocol for cage bioconjugation and analysis. A.F.B.R. synthesized the integrin ligands, established the NMR analysis of the cages, performed most of the encapsulation studies by NMR and wrote the SI. F.R. synthesized the integrin ligands and performed the *in vitro* ELISA tests. M.W. established the functionalisation of peptide **3** and synthesised **L3**. B.S.L. was involved in the synthesis of integrin binding ligands. B.A. performed the *in vitro* experiments. M.N.W. performed some of the encapsulation studies and assisted in the design of the NMR experiments. B.W. optimized the synthesis of ligand **L0**, synthesized cage **C0** and performed NMR experiments. S. S. performed the ICP MS analysis. All the authors contributed to the writing of the manuscript.

References

- 1 T. R. Cook, V. Vajpayee, M. H. Lee, P. J. Stang, K.-W. Chi, *Acc. Chem. Res.* **2013**, *46*, 2464-2474.
- 2 A. Casini, B. Woods, M. Wenzel, *Inorg. Chem.* **2017**, *56*, 14715-14729.
- 3 R. A. Kaner, S. J. Allison, A. D. Faulkner, R. M. Phillips, D. I. Roper, S. L. Shepherd, D. H. Simpson, N. R. Waterfield, P. Scott, *Chem. Sci.* **2016**, *7*, 951-958.
- 4 D. Preston, S. M. McNeill, J. E. M. Lewis, G. I. Giles, J. D. Crowley, *Dalton Trans.* **2016**, *45*, 8050-8060.
- 5 F. Schmitt, N. P. E. Barry, L. Juillerat-Jeanneret, B. Therrien, *Bioorganic Med. Chem. Lett.* **2012**, *22*, 178-180.
- 6 V. Vajpayee, Y. J. Yang, S. C. Kang, H. Kim, I. S. Kim, M. Wang, P. J. Stang, K.-W. Chi, *Chem. Commun.* **2011**, *47*, 5184-5186.

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2
3 7 A. Ahmedova, D. Momekova, M. Yamashina, P. Shestakova, G. Momekov, M. Akita, M.
4 Yoshizawa, *Chem. Asian J.* **2016**, *11*, 474-477.
5
6 8 A. Schmidt, A. Casini, F. E. Kühn, *Coord. Chem. Rev.* **2014**, *275*, 19-36.
7
8 9 T. R. Cook, Y.-R. Zheng, P. J. Stang, *Chem. Rev.* **2013**, *113*, 734-777.
9
10 10 J. E. M. Lewis, E. L. Gavey, S. A. Cameron, J. D. Crowley, *Chem. Sci.* **2012**, *3*, 778-784.
11 11 H. Burger, W. J. Loos, K. Eechoute, J. Verweij, R. H. Mathijssen, E. A. Wiemer, *Drug*
12 *Resist. Updat.* **2011**, *14*, 22-34.
13
14 12 A. Schmidt, M. Hollering, M. Drees, A. Casini, F. E. Kühn, *Dalton Trans.* **2016**, *45*, 8556-
15 8565.
16
17 13 A. Schmidt, M. Hollering, J. Han, A. Casini, F. E. Kühn, *Dalton Trans.* **2016**, *45*, 12297-
18 12300.
19
20 14 F. Kaiser, A. Schmidt, W. Heydenreuter, P. J. Altmann, A. Casini, S. A. Sieber, F. E.
21 Kühn, *Eur. J. Inorg. Chem.* **2016**, *2016*, 5189-5196.
22
23 15 A. Schmidt, V. Molano, M. Hollering, A. Pöthig, A. Casini, F. E. Kühn, *Chem. Eur. J.*
24 **2016**, *22*, 2253-2256.
25
26 16 M. L. Hovlid, N. F. Steinmetz, B. Laufer, J. L. Lau, J. Kuzelka, Q. Wang, T. Hyypiä, G. R.
27 Nemerow, H. Kessler, M. Manchester, M. G. Finn, *Nanoscale* **2012**, *4*, 3698-3705.
28
29 17 J. Han, A. Schmidt, T. Zhang, H. Permentier, G. M. M. Groothuis, R. Bischoff, F. E.
30 Kühn, P. Horvatovich, A. Casini, *Chem. Commun.* **2017**, *53*, 1405-1408.
31
32 18 K. Ley, J. Rivera-Nieves, W. J. Sandborn, S. Shattil, *Nat. Rev. Drug Discovery* **2016**, *15*,
33 173.
34
35 19 M. Nieberler, U. Reuning, F. Reichart, J. Notni, H.-J. Wester, M. Schwaiger, M.
36 Weinmüller, A. Räder, K. Steiger, H. Kessler, *Cancers* **2017**, *9*, 116;
37 doi:10.3390/cancers9090116.
38
39 20 A. Meyer, J. Auernheimer, A. Modlinger, H. Kessler, *Curr. Pharm. Des.* **2006**, *12*, 2723-
40 2747.
41
42 21 D. Heckmann, A. Meyer, L. Marinelli, G. Zahn, R. Stragies, H. Kessler, *Angew. Chem.*
43 *Int. Ed.* **2007**, *46*, 3571-3574; *Angew. Chem.* **2007**, *119*, 3641-3644.
44
45 22 M. Schottelius, B. Laufer, H. Kessler, H.-J. Wester, *Acc. Chem. Res.* **2009**, *42*, 969-980.
46
47 23 F. Rechenmacher, S. Neubauer, J. Polleux, C. Mas-Moruno, M. De Simone, E. A.
48 Cavalcanti-Adam, J. P. Spatz, R. Fässler, H. Kessler, *Angew. Chem. Int. Ed.* **2013**, *52*,
49 1572-1575; *Angew. Chem.* **2013**, *125*, 1612-1616.
50
51 24 S. Neubauer, F. Rechenmacher, A. J. Beer, F. Curnis, K. Pohle, C. D'Alessandria, H.-J.
52 Wester, U. Reuning, A. Corti, M. Schwaiger, H. Kessler, *Angew. Chem. Int. Ed.* **2013**, *52*,
53 11656-11659; *Angew. Chem.* **2013**, *125*, 11870-11873.
54
55
56
57
58
59
60

- 1
2
3 25 T. G. Kapp, F. Rechenmacher, S. Neubauer, O. V. Maltsev, E. A. Cavalcanti-Adam, R.
4 Zarka, U. Reuning, J. Notni, H.-J. Wester, C. Mas-Moruno, J. Spatz, B. Geiger, H.
5 Kessler, *Sci. Rep.* **2017**, *7*, 39805.
6
7
8 26 M. Weinmüller, F. Rechenmacher, U. K. Marelli, F. Reichart, T. G. Kapp, A. F. B. Räder,
9 F. S. Di Leva, L. Marinelli, E. Novellino, J. M. Muñoz-Félix, K. Hodivala-Dilke, A.
10 Schumacher, J. Fanous, C. Gilon, A. Hoffman, H. Kessler, *Angew. Chem. Int. Ed.* **2017**,
11 *56*, 16405-16409; *Angew. Chem.* **2017**, *129*, 16624-16629.
12
13
14 27 M. Aumailley, M. Gurrath, G. Müller, J. Calvete, R. Timpl, H. Kessler, *FEBS Lett.* **1991**,
15 *291*, 50-54.
16
17 28 M. A. Dechantsreiter, E. Planker, B. Mathä, E. Lohof, G. Hölzemann, A. Jonczyk, S. L.
18 Goodman, H. Kessler, *J. Med. Chem.* **1999**, *42*, 3033-3040.
19
20 29 C. Mas-Moruno, F. Rechenmacher, H. Kessler, *Anti-Cancer Agents in Med. Chem.* **2010**,
21 *10*, 753-768.
22
23 30 S. Mukhopadhyay, C. M. Barnés, A. Haskel, S. M. Short, K. R. Barnes, S. J. Lippard,
24 *Bioconjugate Chem.* **2008**, *19*, 39-49.
25
26 31 M. A. Medrano, M. Morais, V. F. C. Ferreira, J. D. G. Correia, A. Paulo, I. Santos, C.
27 Navarro-Ranninger, A. A. Valdes, A. Casini, F. Mendes, A. G. Quiroga, *Eur. J. Inorg.*
28 *Chem.* **2017**, *2017*, 1835-1840.
29
30 32 A. Gandioso, E. Shaili, A. Massaguer, G. Artigas, A. González-Cantó, J. A. Woods, P. J.
31 Sadler, V. Marchán, *Chem. Commun.* **2015**, *51*, 9169-9172.
32
33 33 S. Neubauer, F. Rechenmacher, R. Brimiouille, F. S. Di Leva, A. Bochen, T. R. Sobahi,
34 M. Schottelius, E. Novellino, C. Mas-Moruno, L. Marinelli, H. Kessler, *J. Med. Chem.*
35 **2014**, *57*, 3410-3417.
36
37 34 D. Heckmann, A. Meyer, B. Laufer, G. Zahn, R. Stragies, H. Kessler, *ChemBioChem*
38 **2008**, *9*, 1397-1407.
39
40 35 R. Haubner, R. Gratias, B. Diefenbach, S. L. Goodman, A. Jonczyk, H. Kessler, *J. Am.*
41 *Chem. Soc.* **1996**, *118*, 7461-7472.
42
43 36 C. Mas-Moruno, R. Fraioli, F. Rechenmacher, S. Neubauer, T. G. Kapp, H. Kessler,
44 *Angew. Chem. Int. Ed.* **2016**, *55*, 7048-7067; *Angew. Chem.* **2016**, *128*, 7162-7183.
45
46 37 A. O. Frank, E. Otto, C. Mas-Moruno, H. B. Schiller, L. Marinelli, S. Cosconati, A.
47 Bochen, D. Vossmeier, G. Zahn, R. Stragies, E. Novellino, H. Kessler, *Angew. Chem.*
48 *Int. Ed.* **2010**, *49*, 9278-9281; *Angew. Chem.* **2010**, *122*, 9465-9468.
49
50 38 J. E. Gestwicki, C. W. Cairo, L. E. Strong, K. A. Oetjen, L. L. Kiessling, *J. Am. Chem.*
51 *Soc.* **2002**, *124*, 14922-14933.
52
53
54
55
56
57
58
59
60

- 1
2
3 39 L. L. Kiessling, J. E. Gestwicki, L. E. Strong, *Angew. Chem. Int. Ed.* **2006**, *45*, 2348-
4 2368; *Angew. Chem.* **2006**, *118*, 2408-2429.
5
6 40 G. Thumshirn, U. Hersel, S. L. Goodman, H. Kessler, *Chem. Eur. J.* **2003**, *9*, 2717-2725.
7
8 41 Z.-H. Jin, J. Razkin, V. Josserand, D. Boturyn, A. Grichine, I. Texier, M.-C. Favrot, P.
9 Dumy, J.-L. Coll, *Molecular Imaging* **2007**, *6*, 43-55.
10
11 42 H.-J. Wester, H. Kessler, *J. Nucl. Med.* **2005**, *46*, 1940 -1945.
12
13 43 Y. Zhang, Y. Sun, F. Yang, J. Guo, J. He, Q. Wu, W. Cao, L. Lv, H.
14 Zheng, Z. Zhang, *Viruses* **2013**, *5*, 1114–1130.
15
16 44 M. Pisano, D. E. P. I, V. Nieddu, I. Sassu, S. Cossu, G. Galleri, A. Del Gatto, M. Budroni,
17 A. Cossu, M. Saviano, G. Palmieri, L. Zaccaro, C. Rozzo, *Anticancer Res.* **2013**, *33*, 871-
18 879.
19
20 45 S. L. Goodman, H. J. Grote, C. Wilm, *Biol. Open* **2012**, *1*, 329-340.
21
22 46 I. A. M. de Graaf, P. Olinga, M. H. de Jager, M. T. Merema, R. de Kanter, E. G. van de
23 Kerkhof, G. M. M. Groothuis, *Nat. Protoc.* **2010**, *5*, 1540.
24
25 47 B. Bertrand, L. Stefan, M. Pirrotta, D. Monchaud, E. Bodio, P. Richard, P. Le Gendre, E.
26 Warmerdam, M. H. de Jager, G. M. M. Groothuis, M. Picquet, A. Casini, *Inorg. Chem.*
27 **2014**, *53*, 2296-2303.
28
29 48 S. Daum, V. F. Chekhun, I. N. Todor, N. Y. Lukianova, Y. V. Shvets, L. Sellner, K.
30 Putzker, J. Lewis, T. Zenz, I. A. M. de Graaf, G. M. M. Groothuis, A. Casini, O. Zozulia,
31 F. Hampel, A. Mokhir, *J. Med. Chem.* **2015**, *58*, 2015-2024.
32
33 49 S. Spreckelmeyer, N. Estrada-Ortiz, G. G. H. Prins, M. van der Zee, B. Gammelgaard, S.
34 Stürup, I. A. M. de Graaf, G. M. M. Groothuis, A. Casini, *Metallomics* **2017**, *9*, 1786-1795.
35
36 50 N. Estrada-Ortiz, F. Guarra, I. A. M. de Graaf, L. Marchetti, M. H. de Jager, G. M. M.
37 Groothuis, C. Gabbiani, A. Casini, *ChemMedChem* **2017**, *12*, 1429-1435.
38
39 51 S. Spreckelmeyer, C. Orvig, A. Casini, *Molecules* **2014**, *19*, 15584.
40
41 52 S. A. L. Rousseaux, J. Q. Gong, R. Haver, B. Odell, T. D. W. Claridge, L. M. Herz, H. L.
42 Anderson, *J. Am. Chem. Soc.* **2015**, *137*, 12713-12718.
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

