The X-Ray Structure of the Haloalcohol Dehalogenase HheA from *Arthrobacter* sp. Strain AD2: Insight into Enantioselectivity and Halide Binding in the Haloalcohol Dehalogenase Family

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Haloalcohol dehalogenases are bacterial enzymes that cleave the carbon-halogen bond in short aliphatic vicinal haloalcohols, like 1-chloro-2,3-propanediol, some of which are recalcitrant environmental pollutants. They use a conserved Ser-Tyr-Arg catalytic triad to deprotonate the haloalcohol oxygen, which attacks the halogen-bearing carbon atom, producing an epoxide and a halide ion. Here, we present the X-ray structure of the haloalcohol dehalogenase HheAD2 from *Arthrobacter* sp. strain AD2 at 2.0-Å resolution. Comparison with the previously reported structure of the 34% identical enantioselective haloalcohol dehalogenase Hhec from *Agrobacterium radiobacter* AD1 shows that HheAD2 has a similar quaternary and tertiary structure but a much more open substrate-binding pocket. Docking experiments reveal that HheAD2 can bind both enantiomers of the haloalcohol substrate 1-p-nitrophenyl-2-chloroethanol in a productive way, which explains the low enantiopreference of HheAD2. Other differences are found in the halide-binding site, where the side chain amino group of Asn182 is in a position to stabilize the halogen atom or halide ion in HheAD2, in contrast to Hhec, where a water molecule has taken over this role. These results broaden the insight into the structural determinants that govern reactivity and selectivity in the haloalcohol dehalogenase family.

Dehalogenases are enzymes that cleave carbon-halogen bonds (6). Over the years, detailed structural information has become available for several dehalogenases belonging to evolutionary unrelated families (3). One such family, which was recently characterized, comprises haloalcohol dehalogenases (4, 18, 21). The enzymes from this family were isolated from bacteria that are able to grow on vicinal haloalcohols, like 1,3-dichloro-2-propanol and 2,3-dichloro-1-propanol, or compounds that are degraded via haloalcohols, some of which are notable environmental pollutants. These dehalogenases catalyze the intramolecular substitution of the halogen atom of the haloalcohol function by the neighboring hydroxyl group, thereby producing an epoxide and a halide ion (Fig. 1A). The epoxide product can subsequently be hydrolyzed by an epoxide hydrolase, yielding the corresponding 1,2-diol (14).

Until now, six different haloalcohol dehalogenases, which can be grouped into the three subtypes (A, B, and C) on the basis of amino acid sequence similarities, have been isolated (21). Their sequences also indicated that they all are evolutionarily related to the large family of NAD(P)(H)-dependent short-chain dehydrogenases/reductases (SDRs) (21), which use a similar Ser-Tyr-Lys/Arg catalytic triad to catalyze various alcohol-ketone conversions (11). Crystallographic analysis of the haloalcohol dehalogenase Hhec from *Agrobacterium radiobacter* AD1 revealed that the tyrosine residue of the conserved Ser-Tyr-Arg catalytic triad is in a position to activate the hydroxyl group of the haloalcohol function for nucleophilic attack on the vicinal halogen-bearing carbon atom (Fig. 1B) (4). The halogen atom is bound in a spacious halide-binding site, which also stabilizes the halide product. Structure comparison of the C-type haloalcohol dehalogenase Hhec and the SDRs suggested that the different dehalogenase subtypes likely originated from NADH-dependent, rather than NADPH-dependent, SDR precursors (4). The different subtypes show marked differences in their catalytic behaviors: Hhec is highly enantioselective towards various aliphatic and aromatic substrates (5, 8), whereas the A- and B-type dehalogenases have only a modest enantioselectivity (20, 21).

Until now, structural information on the A- or B-type enzymes has not been available. Here we present the 2.0-Å resolution crystal structure of the A-type haloalcohol dehalogenase HheAD2 from the 3-chloro-1,2-propanediol-utilizing *Arthrobacter* sp. strain AD2. The structure reveals an active site, which, in contrast to Hhec, appears capable of binding both enantiomers of a haloalcohol substrate in a productive conformation. Furthermore, a structure comparison shows that the A- and C-type enzymes use different interactions to stabilize the halogen atom and halide ion during the catalytic cycle.

**MATERIALS AND METHODS**

Haloalcohol dehalogenase HheAD2 was purified as published elsewhere (20, 21). Crystals of HheAD2 were obtained from 2 μl hanging drops consisting of equal amounts of protein solution (8 mg/ml) and well solution containing 22% (wt/vol) polyethylene glycol 8000 as the precipitant, 100 mM MES [2-(N-morpholino)ethanesulfonic acid] buffer, pH 6.5, 100 mM magnesium acetate, and 50 mM sodium formate. Plates of dimensions of 1 by 2 by 0.2 mm³ were observed after half a year.

A diffraction data set was collected at a wavelength of 0.93 Å at beamline...
ID14-2 of the ESRF synchrotron (Grenoble, France). The data were processed using DENZO and SCALEPACK (12). The crystal diffracted to 2.0-Å resolution and had space group P1 with cell dimensions of $a = 79.6$ Å, $b = 75.0$ Å, and $c = 111.5$ Å with $\alpha = 97.1^\circ$, $\beta = 95.4^\circ$, and $\gamma = 115.9^\circ$, with two tetramers (each with a molecular mass of $4 \times 26$ kDa) in the asymmetric unit.

Molecular replacement solutions were obtained with the program AMoRe available in CCP4 (2, 10), using diffraction data between 8- and 4-Å resolution. As a search model, the atomic coordinates of the tetrameric HheC from Arthrobacter sp. strain AD1 (PDB code 1PWX [4]) were used. Rigid body refinement of the solutions gave a starting correlation coefficient of 0.73 and an $R$ factor of 32.2%.

A $\Delta$-weighted 2F$_o$ − F$_c$ difference Fourier map (13) calculated at this stage was used in a combined NCS averaging, density modification, and phase extension procedure using the prime-and-switch method available in the program RESOLVE (19). This improved the overall figure of merit from 0.34 to 0.48 in the final cycle of the density modification procedure, and the resulting $\Delta$-weighted electron density maps showed improved electron density for the amino

FIG. 1. (A) Putative degradation route of 1,3-dichloro-2-propanol in Arthrobacter sp. strain AD1. The epoxide hydrolysis step is catalyzed by epoxide hydrolase (EH), whereas the dehalogenation steps are catalyzed by the haloalcohol dehalogenase HheAD2. (B) Catalytic mechanism of haloalcohol dehalogenases HheC and HheAD2. The hydroxyl function of a haloalcohol substrate is bound by a serine (Ser134 in HheAD2 or Ser132 in HheC) and a tyrosine (Tyr147 in HheAD2 or Tyr145 in HheC). The tyrosine functions as the catalytic base, which activates the hydroxyl group adjacent to the halogen atom for nucleophilic attack on the halogen-bearing carbon atom. The abstracted proton is released to the solvent via the arginine (Arg151 in HheAD2 or Arg149 in HheC), the water molecules, and Asp80 at the surface of both enzymes.
TABLE 1. Data collection and refinement statistics

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<tr>
<th>Parameter</th>
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<td>RMSD B factors (Å) (main chain/ side chain)</td>
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<tr>
<td>Ramachandran plot (%)</td>
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</tr>
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* Abbreviations: a.u., asymmetric unit; RMSD, root mean square deviation; fav., favored; all., allowed; gen. all., generously allowed; disall., disallowed.
* *R*,,, = <S>-<I>, where <S> is the observed intensity and <I> the average intensity, or standard deviation.

The most notable difference between the active sites of HheAAD2 and HheC is the much more open structure of the substrate-binding pocket of HheAAD2 (Fig. 2C and D). This is partly due to the absence of the C-terminal extension in HheAAD2, which precludes the contribution of a residue from a neighboring subunit to the active site, such as Trp249 does in HheC. Another important contribution to the openness of the active site is the presence of a leucine at position 141, whereas HheC has a tryptophan (Trp139) at the equivalent position. Mutation of Trp139 into a phenylalanine previously demonstrated that the voluminous tryptophan side chain is the main determinant of the high enantioselectivity of HheC (16). Manual docking of the R- and S-enantiomers of the haloalcohol substrate 1-p-nitrophenyl-2-chloroethanol in the active site of HheAAD2 shows that both enantiomers can be accommodated without severe clashes (Fig. 2C). This is in agreement with the lack of enantiopreference of HheAAD2 (21).

Halide-binding site of HheA. Like HheC, HheAAD2 contains a spacious halide-binding site. A water molecule is bound in the center of it, occupying a position equivalent to that of the halide ions in the structures of two product complexes of HheC (4). Such an apo structure is in agreement with the absence of halide salts in the crystallization solution of HheAAD2. The halide-binding site is formed by two loops extending from the βBAβBB Rossmann fold motif. The first and largest one is equivalent to a highly variable loop that is responsible for cofactor and substrate binding in members of the evolutionarily related SDR family, whereas the second loop corresponds to the Gly-rich cofactor-binding motif of the SDR enzymes (4, 11). The two loops in HheAAD2 are almost identical to their equivalents in HheC, with the exception of the region from residue 180 to residue 185. The amino acid sequence of this region in HheAAD2 (180-Phe-Asn-Asn-Pro-Thr-Tyr-185) is three residues shorter than that in HheC (178-Leu-His-Ser-Glu-Asp-Ser-Pro-Tyr-Phe-186 [the three additional residues are shown in bold]) and has a different amino acid composition. These sequence differences typify the A- and C-type haloalcohol dehalogenases and have important consequences for the way a bound halogen atom or halide ion is stabilized in the two enzymes.

In HheAAD2, the water molecule in the halide-binding site interacts with the backbone NH group of Phe180 and the side chain amino group of Asn182 (Fig. 3A). The side chain of Asn182 is kept in position by hydrogen bonds to the carbonyl oxygen atom of Phe180 and to the backbone NH groups of Thr184 and Tyr185. These interactions are facilitated by the sharp turn of the loop at Pro183, which directs the backbone NH groups of Thr184 and Tyr185 inwards. In this way the Asn182 side chain is maintained in a conformation favorable for interaction with a bound halide ion. In HheC, Ser180 is the residue equivalent to Asn182 of HheAAD2 but its side chain is ori-
ented away from the halide ion (Fig. 3B), which, instead, interacts with a water molecule. This water molecule is hydrogen bonded to the backbone NH group of Phe\textsubscript{186} of the \textit{cis} peptide between Tyr\textsubscript{185} and Phe\textsubscript{186} and the backbone carbonyl group of Leu\textsubscript{178}, whose atoms are equivalent to the main chain atoms that bind the Asn\textsubscript{182} side chain in Hhe\textsubscript{AD2}. Thus, part of the large halide-binding loop adopts different conformations in Hhe\textsubscript{AD2} and HheC, which results in local differences in the way the A- and C-type haloalcohol dehalogenases stabilize a bound halogen atom or halide ion.

FIG. 2. (A) Tetrameric structure of the haloalcohol dehalogenase Hhe\textsubscript{AD2} from \textit{Arthrobacter} sp. strain AD2. The location of the active site in each monomer is indicated by the catalytic serine, tyrosine, and arginine in stick representation. (B) Close-up view of the superposition of Hhe\textsubscript{AD2} (purple) and HheC (pink), showing the extended C terminus of HheC that contains the side chain of Trp\textsubscript{249} that binds in the substrate-binding pocket of an opposite subunit. The catalytic residues are shown in stick representation. (C) Stereo view of the active site pocket of native Hhe\textsubscript{AD2} containing the manually docked \textit{R}-enantiomer (green) and \textit{S}-enantiomer (brown) of the haloalcohol substrate 1-p-nitrophenyl-2-chloroethanol. Both enantiomers can be accommodated without severe clashes with the enzyme. (D) Stereo view of the active site pocket of the HheC·Cl\textsuperscript{-}·(\textit{R})-styrene oxide (SO) complex (4). Residues, azidoalcohol, and epoxide compounds are shown in ball-and-stick representation, and the chloride ion is shown as a green sphere. Hydrogen bonds are indicated by black dashed lines. Trp\textsubscript{249}* is contributed by a different monomer in the tetramer of HheC.
Recently, it was shown for HheC that disruption of two hydrogen bonds between the side chains of Asn176, Tyr187, and Trp249 increases the rate of halide release and can enhance the overall catalytic activity (17). This suggests that in HheC the halide release pathway is near the side chains of these residues. However, in HheAAD2, the tyrosine is replaced by a phenylalanine and the tryptophan residue is absent, and therefore further research is needed to establish the halide release pathway in HheAAD2.

REFERENCES


