Appendices Bachelor's degree programme Chemistry

Appendix I Learning outcomes of the degree programme (Article 1.3)

A. Generic learning outcomes - Knowledge

- A1. Bachelor's graduates have general knowledge of the foundations and history of mathematics, natural sciences and technology, in particular those of their own discipline.
- A2. Bachelor's graduates have mastered the basic concepts of their own discipline (see Appendix 1 for further specification) to a certain extent and are familiar with the interrelationships of these concepts within their own discipline as well as with other disciplines.
- A3. Bachelor's graduates have in-depth knowledge of several current topics within their own discipline.
- A4. Bachelor's graduates are familiar with the quantitative character of the fields of mathematics and natural sciences and have an understanding of the methods used in these fields, and particularly within their own discipline, including computer-aided methods.
- A5. Bachelor's graduates have sufficient knowledge and understanding of mathematics and natural sciences to successfully complete a follow-up Master's degree programme in their own discipline.
- A6. Bachelor's graduates are aware of the societal, ethical and social aspects involved in the fields of mathematics and natural sciences.

B. Generic learning outcomes - Skills

- B1. (Research) Bachelor's graduates are able to draw up a research question, , plan and conduct research and report on it independently with a certain degree of supervision. Bachelor's graduates are able to evaluate the value and limitations of their research and assess its applicability outside their own field.
- B2. (Gathering information) Bachelor's graduates are able to gather relevant information using modern means of communication and to critically interpret this information.
- B3. (Collaborating) Bachelor's graduates are able to collaborate in teams (including multidisciplinary teams) on technical-scientific problems.
- B4. (Communicating) Bachelor's graduates are able to communicate orally and in writing in academic and professional contexts, with both colleagues and others. They are familiar with the relevant means of communication.
- B5. (Reflecting) Bachelor's graduates are able to assess their own actions and those of others in a natural sciences context, bearing in mind the social/societal and ethical aspects.
- B6. (Learning skills) Bachelor's graduates are able to apply learning skills that enable them to pursue a follow-up degree and acquire knowledge in new fields with a high level of autonomy.
- B7. Additional subject-specific skills are listed in Appendix 2.

C. Degree programme-specific learning outcomes - Basic Knowledge

The Bachelor's graduate in Chemistry has:

1.1. knowledge of the most important fields of Chemistry: Inorganic, Organic, Analytical, Physical, Polymer Chemistry and Biochemistry, furthermore general knowledge of more specific field such as Theoretical Chemistry, Materials Chemistry, etc.,

- 1.2. knowledge of at least one multidisciplinary field: 'Chemistry of Life', 'Smart Materials' and 'Sustainable Energy and Chemistry',
- 1.3. a broad general knowledge of subjects within his/her own discipline or of subjects within a different discipline,
- 1.4. necessary background knowledge of Mathematics and Physics,
- 1.5. understanding of the position and role of the discipline within science and society, and also in the international character of the discipline.

The Bachelor's graduate has become familiar with the following key elements of Chemistry:

- a. The important aspects of chemical terminology, nomenclature and conventions.
- b. Numerical and computational skills, including error analysis, understanding of the proper order of magnitude and correct use of units.
- c. The most important types of chemical reactions and their characteristics.
- d. The principles and procedures that are used in the chemical analysis and in the characterization of chemical compounds.
- e. The fundamental techniques of structural analysis, including spectroscopy.
- f. The properties of various states of matter and the common theories to describe them.
- g. The principles of Quantum Mechanics and its applications in the description of structure and properties of atoms and molecules.
- h. The principles of Thermodynamics and its applications in Chemistry.
- *i.* The kinetics of chemical processes, catalysis and mechanical interpretation of chemical reactions.
- *j.* The typical properties of elements and their compounds, including group relationships and trends in the periodic table.
- *k.* The structural properties of chemical elements and their compounds.
- 1. The typical properties of aliphatic, aromatic, heterocyclic and organometallic compounds.
- *m.* The nature and behavior of functional groups in molecules.
- *n.* Important synthetic routes of organic/inorganic chemistry.
- o. The relationship between bulk properties of matter and properties of individual atoms and molecules, including macromolecules (both natural and synthetic).
- *p.* The structure and reactivity of important types of biomolecules and the chemistry of important biological processes.
- q. The design of processes (also on industrial scale), taking into account flow and transfer of matter and energy.
- r. Properties of chemicals and the involved environmental and safety aspects.

D. Degree programme-specific learning outcomes outcomes - Skills

The Bachelor's graduate in Chemistry has developed the skills and competencies mentioned below.

Chemistry-related cognitive skills and competencies

The Bachelor's graduate is:

- 2.1. able to demonstrate and use his/her knowledge and understanding of essential facts, concepts, principles and theories related to the topics, as defined in Appendix Ia, in various situations,
- 2.2. able to apply knowledge and understanding to solve basic qualitative and quantitative problems,
- 2.3. skilled in evaluating, interpreting and combining chemical information and data,
- 2.4. able to recognize and implement 'good laboratory practice',
- 2.5. familiar with project work,
- 2.6. able to adopt a professional attitude regarding environmental and safety aspects and possible ethical implications in the context of research, education and industry.

Chemistry-related practical skills

The Bachelor's graduate is:

- 2.7. skilled in the use of standard laboratory procedures and in the use of equipment for synthetic and analytical work,
- 2.8. able to verify chemical properties, to observe and measure events or changes, and to systematically archive and document data,
- 2.9. able to interpret data, obtained from observations and measurements, and relate it to the right theories.
- 2.10. able to assess the risks of laboratory procedures and the use of chemicals,

- 2.11. skilled in the safe handling of chemicals, taking into account physical and chemical properties, including the various specific risks of use, and is also able to act adequately in emergency situations in the laboratory.
- 2.12. able to use IT skills appropriate to the chosen specialization.

Appendix II Follow-on Master's degree programmes (Article 1.5)

The Bachelor's degree programme will grant unconditional admission to the following Master's degree programmes at the University of Groningen:

- Chemistry
- o Education and Communication in Mathematics and Natural Sciences
- Energy and Environmental Sciences

The Bachelor's specialization in Chemistry of Life will grant unconditional admission to the following Master's degree programme:

o Molecular Biology and Biotechnology.

Appendix III Majors and Minors in the degree programme (Article 2.1.2)

The programme consists of a Major and a Minor.

- 1) The elective part within the Major is chosen from the following three tracks:
 - a) Chemistry of Life
 - b) Smart Materials
 - c) Sustainable Chemistry and Energy
- 2) The Minor may either be the Minor Chemistry 'Science for Scientists' or may be chosen from the collection of university and faculty Minors.

Appendix IV Course units in the propaedeutic phase

- List of course units; Article 3.1.1
- Course units with one or more practicals; Article 3.2
- Form of examinations; Article 7.4

Course unit	ECTS	Practical
Calculus for Chemistry and Chemical	5	X
Engineering		
Molecules: Structure, Reactivity, and	5	X
Function		
Choice:	5	
■ From Bacteria to Plastic		X
■ Physics Laboratory 1		X
■ Introduction to Mathematics		
Organic Chemistry 1	5	
Practicum Synthesis and Analysis 1	5	X
Physical Chemistry 1	5	X
Biochemistry	5	X
Biochemistry Practicum	5	X
Spectroscopy	5	X
Process and Product Technology	5	
Inorganic Chemistry	5	
First Year Symposium	5	X

Appendix V Course units in the post-propaedeutic phase - List of course units; Article 6.1

- Course units with one or more practicals; Article 6.2 Compulsory order of examinations; Article 7.2 Form of examinations; Article 7.4

Course unit	ECTS	Practical	Prerequisites
Material Properties 1	5		
Organic Chemistry 2	5		
Practicum Synthesis 2	5	x	
Physical Chemistry 2	5	x	
Linear Algebra for Chemistry	5	x	
and Chemical Engineering			
Quantum chemistry	5	x	
Science, Ethics, Technology	5	x	
and Society			
Macromolecular Chemistry	5	X	
Practicum Macromolecular	5	x	
Chemistry			
Track:	25	See individual	See individual tables
Smart Materials		tables	
Sustainable Chemistry and			
Energy			
Chemistry of Life The algorithm of Life	-		
Track practical	5	X	
Choice:			
Chemical Biology			-
Organic and Molecular			'Practical synthesis and analysis 1' or
Inorganic			'Practicum Chemie voor
Chemistry			Levenswetenschappen'
			'Practical Macromolecular Chemistry'
			-
 Polymer Chemistry 			
 Materials Design, 			
experiment			
Bachelor's Research Project	15	x	Passed 150 ECTS of the Bachelor's degree
			programme of Chemistry
Minor	30	See the	See the programme-specific appendices IV
		programme	and V of the Teaching and Examination
		specific	Regulations.
		appendices IV	
		and V of the	
		Teaching and	
		Examination	
		Regulations.	
		regulations.	

Chemistry of Life

Course unit	ECTS	Practical	Prerequisites
Bio-energetics and Metabolism	5		
Recombinant DNA and	5		
Biotechnology			
Chemical Biology	5	x	
(Bio)-catalysis	5	x	
Cellular Chemistry	5	x	

Smart Materials

Course unit	ECTS	Practical	Prerequisites
Soft Molecular Materials	5		
Material Properties 2	5		
Molecular Design	5	x	
Materials Design: Theoretical	5	x	
Methods			
Trends in Polymer Science	5		

Sustainable Chemistry and Energy

Course unit	ECTS	Practical	Prerequisites
Bioenergy and Bioresources	5		
Electrochemistry and Energy	5	X	
Green Chemistry and Technology	5		
(Bio)-catalysis	5	X	
Physical Organic and Photo-	5		
Chemistry			

The course units 'Bio-energetics and Metabolism' and 'Bioenergy and Bioresources' have a large overlap, therefore only one of both course units may be included in an individual's programme.

Minor Chemistry 'Science for Scientists'

The Minor comprises 30 ECTS and is a coherent and deepening package of course units.

Course unit	ECTS	Practical	Prerequisites
Medicinal Chemistry I	5	X	
Medicinal Chemistry II	5	x	
Structural Probes for Solid	5	x	
Materials			
Organic and Molecular	5		
Electronics			
Solar Cells	5	x	
Electives, courses from	0-30	See the	See the
Bachelor's programmes		programme-	programme-
Chemistry, Chemical		specific	specific
Engineering, Biology, Life		appendices IV	appendices IV
Science and Technology,		and V of the	and V of the
Farmacie, (Applied) Physics,		Teaching and	Teaching and

Astronomy, (Applied)	Examination	Examination
Mathematics, Industrial	Regulations.	Regulations.
Engineering and Management,		
which are to be individually		
approved by the Exam		
Committee.		

Appendix VI Entry requirements Article 10.1

A. HBO (university of applied science) propaedeutic certificate

1. The following requirements apply to the entrance examination as defined in Article 7.28.3 of the Act:

Degree programme	Subjects at VWO (pre- university) level	Requirement: Dutch as a Second Language (programme II) for non- native speakers of Dutch
B Biology	wia or wib + na+sk+bio	Yes
B Pharmacy	wia or wib + na+sk	Yes
B Computing Science	wib	
B Artificial Intelligence	wia or wib	
B Physics	wib+na	
B Chemistry	wib+na+sk	
B Astronomy	wib+na	
B Mathematics	wib	
B Chemical Engineering	wib+na+sk	
B Industrial Engineering and Management Science	wib	
B Applied Physics	wib+na	
B Applied Mathematics	wib	

wia = Mathematics A; wib = Mathematics B; na = Physics; sk = Chemistry; bio = Biology

- 2. Non-native speakers of Dutch who wish to be admitted to the Bachelor's degree programmes in Biology, Life Science and Technology, or Pharmacy must also have passed the State Examination in Dutch as a Second Language, Programme II (NT2-II).
- 3. The Faculty Committee for Special Admissions will determine whether deficiencies have been compensated satisfactorily.

B. Foreign qualifications (EEA)

- 1. Any certificate that grants access to a university in a European country will also grant access to Dutch universities.
- 2. The same requirements that also apply to candidates with an HBO (university of applied science) propaedeutic certificate will apply to these candidates in the entrance examination as defined in Article 7.28.3 of the Act (see A).

- 3. Non-native speakers of Dutch who wish to be admitted to the Bachelor's degree programmes in Biology, Life Science and Technology, or Pharmacy must also have passed the State Examination in Dutch as a Second Language, Programme II (NT2-II).
- 4. In addition, candidates are required to be competent in English: an IELTS score of 6.5, a TOEFL score of 580 (paper-based), of 237 (computer-based) or of 92 (internet-based) or equivalent.
- 5. The Faculty Committee for Special Admissions will determine whether deficiencies have been compensated satisfactorily.

C. Foreign qualifications (German)

- 1. German candidates must have a Zeugnis der Allgemeinen Hochschulreife ('Abitur').
- 2. The following requirements apply to the entrance examination as defined in Article 7.28.3 of the Act:

Degree programme	
B Biology	wi (LK or GK) na (LK or GK) sk (LK or GK) bio (LK or GK) (at least one subject at Leistungskurs level)
B Pharmacy B Life Science and Technology B Chemistry B Chemical Engineering	wi (LK or GK) na (LK or GK) sk (LK or GK) (at least one subject at Leistungskurs level)
B Computing Science B Mathematics B Applied Mathematics B Artificial Intelligence	wi (LK)
B Physics B Astronomy B Applied Physics	wi (LK) na (LK or GK)
B Industrial Engineering and Management Science	wi (LK or GK) na (LK or GK) (at least one subject at Leistungskurs level)

wi= Mathematics; na = Physics; sk = Chemistry; bio = Biology

LK = Leistungskurs level; GK = Grundkurs level followed until end of Class 13 or Class 12 (if Gymnasium education lasts 12 years).

- 3. Non-native speakers of Dutch who wish to be admitted to the Bachelor's degree programmes in Biology, Life Science and Technology, or Pharmacy must also have passed the State Examination in Dutch as a Second Language, Programme II (NT2-II).
- 4. The Faculty Committee for Special Admissions will determine whether deficiencies have been compensated satisfactorily.

D. Foreign qualifications (International Baccalaureate)

1. The following requirements apply to the entrance examination as defined in Article 7.28.3 of the Act:

Degree programme	from 2010/2011
B Biology	Biology (SL or HL) Maths (SL or HL) Physics (SL or HL) Chemistry (SL or HL) two of these subjects at HL
B Pharmacy B Life Science and Technology B Chemistry B Chemical Engineering	Maths (SL or HL) Physics (SL or HL) Chemistry (SL or HL) two of these subjects at HL
B Computing Science B Mathematics B Applied Mathematics	Maths HL
B Artificial Intelligence	Maths SL or Maths HL
B Physics B Astronomy B Applied Physics B Industrial Engineering and Management Science	Maths HL Physics HL

SL = Standard Level, HL = Higher Level

- 2. Non-native speakers of Dutch who wish to be admitted to the Bachelor's degree programmes in Biology, Life Science and Technology, or Pharmacy must also have passed the State Examination in Dutch as a Second Language, Programme II (NT2-II).
- 3. The Faculty Committee for Special Admissions will determine whether deficiencies have been compensated satisfactorily.

E. Foreign qualifications (non-EEA)

- 1. A non-European certificate that according to NUFFIC and/or NARIC standards is equivalent to a Dutch VWO certificate will grant access to university in the Netherlands.
- 2. The same requirements that also apply to candidates with an HBO (university of applied science) propaedeutic certificate will apply to these candidates in the entrance examination as defined in Article 7.28.3 of the Act (see A).
- 3. Non-native speakers of Dutch who wish to be admitted to the Bachelor's degree programmes in Biology, Life Science and Technology, or Pharmacy must also have passed the State Examination in Dutch as a Second Language, Programme II (NT2-II).
- 4. In addition, candidates are required to be competent in English: an IELTS score of 6.5, a TOEFL score of 580 (paper-based), of 237 (computer-based) or of 92 (internet-based) or equivalent.
- 5. The Faculty Committee for Special Admissions will determine whether deficiencies have been compensated satisfactorily.

F. Entrance examination

1. The following requirements apply to the entrance examination as defined in Article 7.29 of the Act:

Degree programme	Nature and Health VWO level	or	Nature and Technology VWO level
B Biology	en, wia or b, sk, bio, na		en, wib, na, sk, bio
B Pharmacy	en, wia or b, sk, bio, na		en, wib, na, sk
B Life Science and Technology	en, wib, sk, bio, na		en, wib, na, sk
B Computing Science	en, wib, sk, bio		en, wib, na, sk
B Artificial Intelligence	en, wia of b, sk, bio		en, wib, na, sk
B Physics	en, wib, sk, bio, na		en, wib, na, sk
B Chemistry	en, wib, sk, bio, na		en, wib, na, sk
B Astronomy	en, wib, sk, bio, na		en, wib, na, sk
B Mathematics	en, wib, sk, bio		en, wib, na, sk
B Chemical Engineering	en, wib, sk, bio, na		en, wib, na, sk
B Industrial Engineering and	en, wib, sk, bio		en, wib, na, sk
Management Science			
B Applied Physics	en, wib, sk, bio, na		en, wib, na, sk
B Applied Mathematics	en, wib, sk, bio		en, wib, na, sk

en = English; wia = Mathematics A; wib = Mathematics B; na = Physics; sk = Chemistry; bio = Biology

- 2. Non-native speakers of Dutch who wish to be admitted to the Bachelor's degree programmes in Biology, Life Science and Technology, or Pharmacy must also have passed the State Examination in Dutch as a Second Language, Programme II (NT2-II).
- 1. The Faculty Committee for Special Admissions will determine whether deficiencies have been compensated satisfactorily.

Appendix VII Clustering of Bachelor's degree programmes Article 4.3.6, Article 4.4.2

Degree programme CROHO code	Name of degree programme	Clustered with CROHO code	Name of degree programme
56286	B Life Science and Technology	56860 56157	B Biology B Pharmacy
56860	B Biology	56286 56157	B Life Science and Technology B Pharmacy
56157	B Pharmacy	56860 56286	B Biology B Life Science and Technology
56980	B Mathematics	56965	B Applied Mathematics
56965	B Applied Mathematics	56980	B Mathematics
50206	B Physics	56962 50205	B Applied Physics B Astronomy
56962	B Applied Physics	50206 50205	B Physics B Astronomy
50205	B Astronomy	56962 50206	B Applied Physics B Physics
56857	B Chemistry	56960	B Chemical Engineering
56960	B Chemical Engineering	56857	B Chemistry

Appendix VIII Admission to the post-propaedeutic phase

The following candidates will be admitted to the post-propaedeutic phase:

 Holders of a propaedeutic certificate of the degree programmes in Chemistry or Chemical Engineering

Appendix IX Contact Hours in the propaedeutic phase

Bachelor year 1	
Type of contact	Number of contact hours per year
Lectures	250
Tutorials	230
Practical	330
Computer practical	90
Study support/Mentor groups	8
Internship support and guidance	-
Exams	52
Misc. contact hours	-